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U.S. ARMY CORPS OF ENGINEERS

Public Hearing
Clean Water Act 401(a) (2) Decision
PolyMet Mining/ NorthMet Mine

DAY 2
May 4, 2022
Black Bear Casino
Carlton, Minnesota
9:00 a.m.

REPORTED BY: Lisa M. Thorsgaard, RPR

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P-R-O-C-E-E-D-I-N-G-S

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3 COLONEL JANSEN: Good morning,
4 everyone, and welcome back for Day 2 of our
5 three-day public hearing regarding the Corps
6 Section 404 permit for the PolyMet NorthMet mine
7 project near Babbitt, Minnesota.

8 Again, my name is Colonel Karl Jansen. I
9 serve as the Commander of the U.S. Army Corps of
10 Engineers, St. Paul District. I thank all parties
11 for joining us again today to provide statements.

12 Our schedule for today is as follows: First,
13 we'll hear PolyMet's views, opinions, and
14 recommendations, and they have a two-hour
15 allocation for this purpose.

16 Following PolyMet's presentation, we'll recess
17 for lunch until 12:30. After lunch, Fond du Lac is
18 allocated a two-hour rebuttal opportunity.
19 Following their rebuttal, we'll take a recess and
20 then PolyMet will have a two-hour rebuttal
21 opportunity as well.

22 Following the rebuttals, I'll review
23 instructions for tomorrow's public hearing and then
24 we'll adjourn for the day.

25 So with that, I'd like to recognize

1 Ms. Christie Kearney.

2 MS. CHRISTIE KEARNEY: Thank you,
3 Colonel, for having us today.

4 My name is Christie Kearney. Kearney is
5 K-E-A-R-N-E-Y. I'm the vice president of
6 environmental affairs for PolyMet Mining. I've
7 been with PolyMet for six years now. I joined the
8 team just before we started environmental
9 permitting, although I was a consultant for PolyMet
10 for 10 years prior to that throughout the
11 environmental review process.

12 My background is I'm a licensed professional
13 environmental engineer. I've been doing
14 environmental review and permitting for close to 25
15 years throughout the Midwest and in Alaska.

16 I'm originally from Hibbing, Minnesota, which
17 is where I live right now with my husband and my
18 daughter. I'm an avid outdoorswoman. I hunt. I
19 fish. I play in northeastern Minnesota. Most of
20 our team live right around the mine and the plant
21 site, and it's important to all of us that we
22 protect the natural resources around us.

23 Yesterday you heard close to 8 hours of
24 testimony from the Band and the Band's experts that
25 we'll violate the Band's water quality standards.

1 This is in direct conflict with our final EIS and
2 our environmental permits.

3 There are two main reasons that we will not be
4 violating the Band's water quality standards.

5 First and foremost is we're cleaning up a
6 Legacy brownfield plant site with Legacy water
7 quality issues which in turn will also clean up the
8 St. Louis River.

9 And second, we're 116 river miles upstream
10 from the Fond du Lac Reservation. The water from
11 our discharges is about 0.5 percent of the flow at
12 Fond du Lac.

13 This is a picture of our plant site, our
14 Legacy plant site. This is the concentrator at the
15 plant. This was built in the mid-1950s. Our
16 tailings basin is shown here in the background, the
17 large green field-looking area and the pond on the
18 right.

19 Today I'm going to start with an overview of
20 our project explaining a little bit about our
21 project, our project's location, where we are in
22 relation to the Fond du Lac Reservation, which you
23 also heard about yesterday, and characterize our
24 discharge relative to the St. Louis River flows.

25 First, let's talk about why we're all here and

1 why this all matters. Our project will mine and
2 process key metals such as copper, nickel, and
3 cobalt which are essential for the clean energy
4 transition. If we assume all of our metals go to
5 electric vehicles, our 20-year mine plan has enough
6 copper to supply 6.7 million electric vehicles,
7 enough nickel to supply 2.6 million electric
8 vehicles, and enough cobalt to supply close to of
9 690,000 electric vehicles.

10 By replacing this number of gas-powered
11 passenger vehicles, we'll be offsetting more than
12 30 million metric tons of CO2 with copper alone.

13 If our metals went to wind turbines instead,
14 we'll have enough copper for approximately 130,000
15 wind turbines.

16 This statement by the World Bank Association
17 in international policy and financing organization
18 is stunning. In the last 5,000 years, about 550
19 millions tons of copper has been produced. The
20 world will need the same amount of copper in the
21 next 25 years to meet the global demand. And this
22 demand is driven by the Clean Energy movement and
23 renewable energy.

24 The Biden Administration has focused on the
25 transition to electric vehicles and renewable

1 energy and has taken many steps over the last year
2 to strengthen and boost the domestic supply chains
3 of critical metals needed.

4 In June 2021, last year, this White House
5 report shown on the left came out as a result of an
6 executive order requesting review of America's
7 supply chains. The PolyMet project was
8 specifically cited on page 99 of this report as a
9 fully-permitted domestic nickel mine.

10 At the end of March, about a month ago,
11 President Biden invoked the Defense Production Act
12 meant to encourage and help responsibly developed
13 projects such as the NorthMet project move forward.

14 Now, let's look at where PolyMet is located in
15 northeastern Minnesota. This figure outlines the
16 St. Louis River watershed, as you've seen
17 yesterday. The stars here show the location of the
18 plant site and mine site at the upper part of the
19 watershed on the eastern side of the Mesabi Iron
20 Range where mining has occurred for over 130 years.
21 Lake Superior is shown at the bottom of the
22 watershed near Fond du Lac's Reservation.

23 There's a few important points here. Our
24 NorthMet project is located at the very top of the
25 watershed, very close to the headwaters of the

1 St. Louis River. You can see the magnitude of the
2 watershed above Fond du Lac's Reservation with
3 nearly 85 percent of the watershed coming in above
4 Fond du Lac and the last major watershed, the
5 Cloquet River, coming in between Fond du Lac's
6 borders.

7 Our nearest discharge point is 116 river miles
8 from Fond du Lac's Reservation. That's from the
9 northern most part of Fond du Lac's Reservation to
10 our closest discharge point at PolyMet.

11 Just to put into context how far Fond du Lac
12 is from the PolyMet project, we have some
13 comparisons shown here. It's 150 road miles from
14 Duluth to St. Paul, 120 road miles from L.A. to San
15 Diego, 116 river miles from PolyMet to Fond du Lac,
16 as I mentioned. 124 miles gets you halfway to the
17 International Space Station. The Corps of
18 Engineers' headquarters is 130 miles from Black
19 Bear Casino. It's a long way.

20 Now I want to call your attention to two
21 specific evaluation locations that we use to
22 bookend potential impacts to Fond du Lac as part of
23 our permitting process. These locations have been
24 used by the DNR and the MPCA in their mercury
25 evaluation reports over the years which is why they

1 were chosen for our project. There's a lot of
2 published literature on mercury specifically from
3 both of these locations.

4 PolyMet and the MPCA reference these two
5 evaluation locations throughout our permit
6 documents, including the 401 water quality
7 certification fact sheet and the NPDES permit fact
8 sheet. This includes the Forbes USGS site, which
9 is 50 miles downstream from PolyMet and 66 miles
10 upstream from Fond du Lac, and the Cloquet River
11 which is 143 miles downstream from PolyMet and just
12 five miles downstream from Fond du Lac.

13 Now, to zoom into these two locations, these
14 two aerial photos are set at the same scale showing
15 the size of the St. Louis River at these two
16 evaluation locations. The average flow at Forbes
17 is 570 CFS. The average flow at Cloquet, it's
18 actually Scanlon just immediately south of Cloquet,
19 is 2300 CFS, four times the flow.

20 This graphic is intended to provide some
21 context to the river flows that we're evaluating.
22 The figure is intended to be to scale relative to
23 the flows of the river. To orient you, our
24 northern streams are shown to the left with the
25 Partridge River, the St. Louis River headquarters,

1 and the Embarrass River shown here. Fond du Lac is
2 shown to the right with the Cloquet River coming in
3 between its borders.

4 Now to add the mine site and the plant site
5 flows. These flows couldn't be shown to scale or
6 you wouldn't see them on the figure.

7 So these are the flows that come from our mine
8 site and plant site. We have 4 CFS coming off of
9 the mine site. This is water that's unimpacted by
10 mining activities, so it's storm water and natural
11 runoff. From the plant site we have approximately
12 8 CFS of flow; 1 CFS going to the Partridge and 7
13 CFS going to the Embarrass River. This is water
14 mainly coming from our wastewater treatment system
15 discharge, so treated wastewater and some storm
16 water.

17 We have evaluation locations just downstream
18 of both our plant site and mine site. Our flows at
19 the mine site are captured in the Partridge River
20 at this location shown here with about 49 CFS of
21 flow, and downstream of the Embarrass -- in the
22 Embarrass River is about 87 CFS of flow.

23 As I mentioned earlier, Forbes is 570 CFS of
24 flow, and the Cloquet -- at Cloquet we have about
25 2300 CFS of flow.

1 To put into context the size of our flows, we
2 have some graphics shown here. Our flow at the
3 mine site is about 10 percent of the flow in the
4 Partridge River. Our flow in the Embarrass River
5 is about 8 percent of the flow in the Embarrass
6 River. At Forbes our discharges from the mine site
7 and plant site represent 2 percent of the flow of
8 the river there. And just downstream of Fond du
9 Lac our mine site plant site flows represents about
10 half a percent of the flow in the river in that
11 location.

12 Now, turning back to the plant site, we saw
13 this figure before, our brownfield plant site.
14 We're going to talk a little bit about our existing
15 conditions that are in place today before the
16 NorthMet project comes online.

17 As I mentioned earlier, we have -- we are
18 using a Legacy taconite mine that has water quality
19 issues on site. Zooming out to an overview of the
20 site, our plant site is shown here. We have
21 streams to the north that are fed from the seeps
22 that are currently coming out of our tailings
23 basin. We have seeps to the south feeding second
24 creek downstream of the plant site. To the east we
25 have our mine site which is the green field with

1 the Partridge River running to the north, to the
2 east, and to the south of it. We're immediately
3 south of the north shore mine, Peter-Mitchell mine.
4 We have a transportation utility corridor between
5 the mine site and plant site with an existing haul
6 road, an existing railroad, and power lines between
7 that we'll be reusing for our project site.

8 And then the last thing I want to point out is
9 Colby Lake which is just south of the plant site.
10 This is a makeup water source for the plant site so
11 when we need additional water to run our plant, we
12 have an appropriation permit to take water from
13 Colby Lake. This is significant because Colby Lake
14 water -- Colby Lake is high in mercury as well.

15 This is an aerial photo zoomed back to the
16 plant site and the tailings basin. So we're
17 reusing a former taconite tailings basin to hold
18 our tailings as well as the plant site which you
19 can see here on the south side of the tailings
20 basin. This large building which was shown in
21 prior photos is about a quarter mile long for
22 context. The tailings basin is very large. It's
23 about four and a half square miles.

24 The tailings basin holds just over 800 million
25 cubic yards of taconite tailings which is the cause

1 of the Legacy water quality problems that we see
2 downstream and around our site. This site has been
3 closed for over 20 years. It closed in January of
4 2021 [sic]. However, the tailings basin is covered
5 under a consent decree and an MPDS permit. It's
6 the source of several elevated constituents to
7 downstream waters including sulfate which ranges
8 from 200 to 300 mg/L and specific conductance which
9 ranges from 900 to 2600 $\mu\text{S}/\text{cm}$ that's currently
10 flowing downstream.

11 The PolyMet's NorthMet project design accounts
12 for these Legacy water quality issues at this
13 brownfield site. It's the water management that
14 we're planning for our project that the Band omits
15 from their "will affect" letter and from their
16 descriptions of our project site.

17 In addition to the water collection and reuse,
18 we'll be using best available technology for water
19 treatment for our project site, membrane treatment
20 technology.

21 So you'll recall this figure showing our mine
22 site and plant site. Zooming over to the mine
23 site, I want to walk you through our project.

24 This figure reflects our maximum build-out
25 approximately mine year 11. The gray polygons here

1 are our mine pits. The yellow polygons are our
2 stockpiles. We have our haul roads shown here in
3 black crosshatch and our overburden storage and
4 laydown area where we're storing peat for future
5 reuse and reclamation shown here.

6 The pink polygons are what I want to call your
7 attention to. These pink polygons are our mine
8 water ponds. We'll be collecting mine water that's
9 impacted from runoff from our stockpiles, our haul
10 roads, our pits, and our overburden storage and
11 laydown area. This water will be pumped down to
12 our equalization basin south of the mine site and
13 pumped over to the plant site for treatment and
14 reuse at the plant site.

15 Separate from our mine water management we
16 have storm water management. This is the natural
17 runoff in storm water that's unimpacted by mining
18 activities. The yellow dashed lines are storm
19 water ditches that route storm water around and
20 away from our mine features. These lead to our
21 blue storm water ponds.

22 The blue ponds are retention ponds that allow
23 storm water to slow and settle any suspended solids
24 from the runoff in storm water before leaving the
25 site.

1 Our mine water management actions collecting
2 water that's been contacted by mine water are
3 estimated to reduce flows on the mine site by about
4 48 percent. This sounds significant but we're just
5 talking about this immediate area.

6 The 48 percent reduction of flows from
7 collection of our water off our mine features is
8 estimated to result in less than a 5 percent change
9 in the Partridge River just south of our project
10 site.

11 We're required by our permit conditions to
12 maintain plus or minus 20 percent of the flows in
13 the Partridge River. Regardless, the mine impacted
14 water capture system is what provides us with a
15 reduction of sulfate and mercury from the mine
16 site.

17 Now to move over to the plant site and
18 tailings basin. We'll use the eastern half of the
19 existing tailings basin for storage of our tailings
20 on the project. This figure is a representation of
21 mine year 20 with full build-out at the end of our
22 project and a pond at the top of our tailings
23 basin.

24 The plant site is shown to the south of the
25 tailings basin with storm water ponds that are

1 fairly small and may be hard to see but they're
2 blue, similar to what you saw at our mine site.
3 And then we have our hydrometallurgical residue
4 facility which is a double line separate system.

5 The important mine water feature I want to
6 call your attention to here is this light blue and
7 dark blue dash line around the tailings basin.
8 This is our seepage containment system. It isn't
9 shown in areas where there's high bedrock where
10 there would be no seepage but it is surrounding our
11 tailings basin.

12 You'll also see arrows coming out from that
13 system. They actually continue along the east side
14 as well. This represents our stream augmentation
15 system. Because we're collecting the seepage
16 that's currently feeding streams to the north and
17 to the south of our tailings basin, we're required
18 by our permits to augment the streams on the
19 outside of our containment system with treated
20 water from our wastewater treatment system. We're
21 required to augment these streams at a rate of plus
22 or minus 20 percent, mimicking natural conditions
23 in the streams. This figure shows that containment
24 wall and cross-section.

25 Seepage flows down through and out of our

1 tailings basin into the surficial aquifer. This is
2 happening now with that water flowing off site and
3 into the downstream waters and ultimately to the
4 St. Louis River.

5 Prior to discharge of any tailings into our
6 tailings basin, we'll be installing a cut-off wall
7 tied into bedrock to stop further seepage from
8 leaving the tailings basin.

9 Nancy mentioned in her presentation yesterday
10 that she's seen these cut-off walls only 50 to
11 60 percent effective. Cut-off walls have been used
12 for decades around the world in landfills, in
13 remediation sites, and in dams by the Corps of
14 Engineers.

15 That seepage stopped from our cut-off wall
16 will be collected in a series of pipes and pumps
17 that will be pumped back up to the tailings basin
18 for future -- for reuse as well as to our water
19 treatment system to clean up and discharge.

20 The wastewater treatment system will -- the
21 augmentation system will be fed by the water
22 treatment system discharged downstream of that
23 cut-off wall system.

24 PolyMet's water management system and
25 treatment system is critical to understand to

1 really understand the protections that are in place
2 for the downstream waters. It's these water
3 management actions that weren't mentioned by the
4 Band yesterday or in their "will affect" letter.

5 Just to walk through them again, PolyMet will
6 collect and treat tailings basin seepage with this
7 containment system. We'll also be collecting and
8 treating mine impacted waters from the mine site.
9 We'll be using water from Colby Lake for plant
10 makeup. Colby Lake being high in mercury, we'll
11 remove that load from the St. Louis River system
12 for use in our plant site and treatment before
13 discharge.

14 These actions will reduce mercury loading,
15 specific conductance, and sulfate loading in the
16 St. Louis River watershed. Sulfate loading will be
17 reduced by 1,380,000 kilograms per year, totaling
18 just under 28 billion kilograms over the 20-year
19 life of the mine.

20 Critical to this reduction is our use of best
21 available technology for treatment of sulfate and
22 other constituents. We're using membrane treatment
23 technology. This water treatment design is a tried
24 and true method used for drinking water around the
25 world and in many mine applications.

1 This technology is actually also used in
2 Michigan over at the Eagle Mine to meet their
3 mercury standard of 1.3 ng/L. Eagle's last permit
4 fact sheet from 2015 states they're required to use
5 a detection limit of 0.5 ng/L of mercury, and
6 they've been measuring nondetects in their
7 discharge, so something less than 0.5 ng/L.

8 We've tested our wastewater treatment system
9 in a pilot plant shown here that ran over
10 3 million gallons through, successfully meeting
11 that 10 mg/L sulfate standard. This is proven
12 technology.

13 The NorthMet project is the only project
14 permitted to meet the sulfate limit of 10 mg/L at
15 the end of the pipe. This was just confirmed with
16 the MPCA last week.

17 The water quality standard is actually at the
18 wild rice stand, and the nearest stand from our
19 discharge is 10 miles downstream. We agreed to
20 meet this 10 mg/L wild rice standard at the end of
21 our pipe during our environmental review process.
22 For comparison, the federal drinking water standard
23 is 250 mg/L.

24 This slide provides some context to the
25 mercury story. Rain is falling on our site at

1 about 11 to 12 ng/L. Runoff in the watershed
2 around our site is 3 1/2 to 6 ng/L. Our permits
3 require us to discharge water at something less
4 than the 1.3 ng/L standard. This is nine times
5 cleaner than rainwater. It's three times cleaner
6 than what's found in the natural watershed.

7 From a simple mass balance perspective, this
8 is easy math. PolyMet's treating runoff from 3 1/2
9 to 6 ng/L from 4800 acres down to something less
10 than 1.3 ng/L. We're removing a lot of mercury.

11 Let's go back to our graphic with these flows.
12 You've seen these numbers already. Our mine water
13 management actions at the mine site will result in
14 a reduction of 4.4 grams per year of mercury from
15 the mine site. So a capture of the 48 percent of
16 the flows at the mine site, which is mostly
17 rainwater, will reduce mercury in the Partridge
18 River by 4.4 grams per year.

19 Our mine water actions at the plant site
20 reduces mercury by 0.8 grams per year in the
21 Embarrass River. At Forbes this equates to a
22 reduction of 5.2 grams per year of mercury from the
23 St. Louis River which is carried down to Cloquet
24 and to the Fond du Lac Reservation.

25 The story is similar but more astounding for

1 sulfate. A collection of rainwater at the mine
2 site, our mine impacted waters, we're reducing
3 sulfate by 100,000 kilograms.

4 At the plant site our seepage containment
5 system which is currently capturing that seepage
6 water that's ranging from 200 to 300 mg/L and we're
7 discharging water at 10 mg/L, results in a
8 significant reduction of sulfate, 1,280,000
9 kilograms per year. Add these together and you get
10 a reduction of about 1.4 million kilograms per year
11 at Forbes which continues down to Cloquet.

12 Over the 20-year life of our project, this
13 results in a little over 100 grams of mercury being
14 removed from the St. Louis River and close to
15 28 billion kilograms of sulfate being removed from
16 the system, all as a result of a brownfield
17 cleanup.

18 It's these reductions that allowed the MPCA to
19 issue the 401 certification.

20 So in summary, our project's water management
21 strategy improves water quality in the St. Louis
22 River. The project's water management actions are
23 designed to remove mercury and sulfate and specific
24 conductance. Most mercury load comes from
25 rainwater which we're collecting and treating. Our

1 wastewater treatment system is best available
2 technology, membrane treatment, to meet that
3 sulfate standard.

4 And the overall design of our project results
5 in reductions of mercury and sulfate loads and
6 specific conductance concentrations in the
7 St. Louis River.

8 Now to look at the Band's claims. Their four
9 main claims in violation of water quality are
10 related to sulfate, mercury, methylmercury, and
11 specific conductance.

12 In making those claims, the Band ignores the
13 project's water management actions. They weren't
14 even mentioned yesterday.

15 They also assert a number of other violations
16 in their "will affect" letter including
17 antidegradation, narrative standards, designated
18 uses, and wetland water quality, but they're all
19 based on a claim that we will significantly
20 increase sulfate, mercury, methylmercury, and
21 specific conductance enough to violate those other
22 water quality standards.

23 We will now have our expert witnesses come up
24 and talk through the Band's claims to show that not
25 only will we not increase the concentrations of any

1 of these parameters, but we will reduce the loads
2 of mercury and sulfate and the concentration of
3 specific conductance in the St. Louis River waters,
4 including at Fond du Lac's Reservation.

5 As a preview, we'll have three different
6 technical experts come up to present the science as
7 it pertains to the Band's claims.

8 Steve Donohue from Foth will come up first who
9 will show how the primary source of mercury to the
10 watershed is from precipitation. He will present
11 the results of a new analysis showing that the
12 project will not cause a measurable change to
13 specific conductance or salinity. Fond du Lac's
14 specific conductance water quality standard of
15 300 $\mu\text{S}/\text{cm}$ was not in effect when our project was
16 permitted, so this is a new analysis.

17 He will also explain the relationship between
18 sulfate, mercury, and methylmercury which you also
19 heard yesterday.

20 Cliff Twaroski from Barr Engineering is our
21 mercury expert that led our cross-media analysis
22 and permitting and the mercury work in our EIS
23 process. Cliff will summarize the detailed water
24 modeling work that shows the project will decrease
25 the loading of sulfate, mercury, and methylmercury

1 in the St. Louis River.

2 And lastly, we'll have Greg Council from Tetra
3 Tech who will explain how the Band's assertions of
4 indirect wetland impacts are significantly
5 overstated, explain why the methods they state
6 should have been used are not actually appropriate,
7 and how the processes they state will cause
8 concerns actually result in less methylmercury
9 reaching the St. Louis River.

10 MR. STEVE DONOHUE: Good morning,
11 Colonel Jansen and participants in this hearing.
12 My name is Steve Donohue. Last name spelled
13 D-O-N-O-H-U-E. I work at Foth Infrastructure &
14 Environment, and I am the vice president of mining
15 sector services.

16 I have, by way of background, about 32 years
17 of experience working in the mining industry,
18 principally on the permitting of new metallic
19 mining projects, including the Eagle project that
20 Christie Kearney referenced previously. I'm also a
21 licensed professional hydrologist, so my technical
22 expertise is in the area of water resources,
23 hydrology, and hydrogeology.

24 I'm going to focus today on four key points
25 that we actually addressed in technical memoranda

1 that have been provided to the U.S. Army Corps of
2 Engineers.

3 First, we're going to focus on mercury loading
4 to the St. Louis River and provide information to
5 the hearing here showing how that is driven by
6 atmospheric processes, principally precipitation
7 that deposits mercury into the entirety of the
8 St. Louis River watershed and also the same type of
9 process occurs with sulfate. That is the driver
10 behind the behavior that we see in the St. Louis
11 River related to mercury as it relates to the Fond
12 du Lac Reservation and the St. Louis River as it
13 flows by and adjacent to that reservation.

14 Secondly, we're going to focus on another
15 memoranda that we provided which is the Band's
16 claim that specific conductance will be violated in
17 the waters of the Band. The analysis that we
18 provided in the memorandum looks at the various
19 types of discharges that are going to take place,
20 the various types of water management activities
21 that are going to take place on the project that
22 are going to pull out things like sulfate and other
23 constituents and show how that reduces, actually,
24 specific conductance in the St. Louis River and
25 also reduces salinity in the St. Louis River which

1 is another issue that has been raised by the Band.

2 Finally, I will focus on water level
3 fluctuations, the inputs of sulfate and mercury via
4 these atmospheric processes into the wetlands, and
5 the types of behavior that occurs in these wetlands
6 that drives the generation of methylmercury which,
7 as we heard from Dr. Branfireun yesterday, is the
8 form of mercury that people are concerned about as
9 it relates to migration through these water
10 resources.

11 So to begin with, we're going to start at a
12 fairly high level here, and what we have here is
13 we're looking at mercury loadings via precipitation
14 upstream of the Fond du Lac Reservation boundary.
15 This is based off of data that is referenced in
16 this footnote here on the figure where we've looked
17 at data on what is the mercury concentration in
18 precipitation in this area. And on average it
19 averages about 11.7 ng/L in precipitation over the
20 last 20 years.

21 If we apply that precip, which is about 29.8
22 inches per year, at 11.7 ng in that precip water to
23 these various watersheds, sub-watersheds that drain
24 into the St. Louis River, and we've identified all
25 these sub-watersheds along the bottom of the graph

1 here, we can see that some of these sub-watersheds
2 are catching significant amounts of mercury just
3 due to precipitation alone. That is what some of
4 that mercury, although it all doesn't run off into
5 the streams that feed the St. Louis River
6 watershed, it is the driver behind the mercury that
7 ends up there. It is the driver behind the mercury
8 that ends up in the wetlands where some of it is
9 converted to methylmercury which then makes its way
10 into the river that flows by the Reservation.

11 The least significant source of natural
12 mercury input into this watershed is the
13 sub-watershed around the NorthMet project. This
14 figure alone demonstrates that the behavior of
15 mercury in the St. Louis River near the Reservation
16 is really driven by these other watersheds and
17 what's occurring there naturally via precipitation.

18 As Christie Kearney mentioned, once the
19 project goes into operation, water management is
20 key to the behavior of mercury as it relates to the
21 PolyMet project. With that capture system around
22 the flow -- the tailings basin, with the
23 containment of water at the mine site, the pumping
24 of that water through the water treatment system,
25 we will actually see a reduction in mercury loading

1 in the sub-watershed around the project by about
2 5.2 ng/L. It's all driven by the engineering and
3 the treatment of water at the mine operation.

4 Another way of looking at this is on this flow
5 diagram that Christie provided earlier. Here we've
6 got all the different tributaries flowing into the
7 St. Louis River. We've got the mine site tributary
8 and the plant site tributaries, which are pretty
9 minor, flowing into the Partridge River and the
10 Embarrass River which feed the St. Louis River.

11 We're looking at that same data that we had on
12 the previous figure. This is -- if we look at the
13 upper part of the Partridge River, this is the
14 amount of mercury that's coming into that watershed
15 vis-a-vis precipitation on an annual basis.

16 Same thing in the Embarrass River. We've got
17 about 4,183 g/yr that's impinging on that watershed
18 on an annual basis. A portion of that ends up in
19 the river that then drains into the St. Louis
20 River.

21 Likewise, as we go downstream and we see
22 larger watersheds feeding the St. Louis River, that
23 loading of mercury and as a result methylmercury
24 that goes into the river increases such that by the
25 time we get to the upstream boundary of the

1 Reservation there's about 56,000 g/yr of mercury
2 that is naturally coming into the system in this
3 watershed, some of which makes its way into the
4 St. Louis River and drives the mercury behavior
5 that we see in the St. Louis River.

6 Another way of looking at it is vis-a-vis this
7 pie chart. We're looking at the same type of data.
8 And we can see that the NorthMet project under
9 existing conditions is an insignificant contributor
10 to the amount of mercury loading that takes place
11 in this watershed on an annual basis.

12 Again, once the project goes into operation,
13 the amount of mercury coming from the small
14 watershed around the PolyMet project, which is
15 already a small contributor to this watershed, is
16 actually going to be reduced. Again, it's due to
17 the containment of water that capture that water
18 and the treatment of that water through the
19 membrane filtration technology that's built into
20 the water treatment system.

21 As Christie mentioned, we know that this
22 technology works. It's not speculative. It's been
23 used at other mine sites, notably the Eagle mine
24 site. And just by way of background, there were a
25 lot of claims made at that project 10 years ago

1 that the technology wouldn't work. It has worked.
2 It's in operation and it's working today.

3 Here we're looking at the same data with the
4 project in operation. And again, about a 5.2 gram
5 of mercury per year reduction that's going to go
6 into the St. Louis River from the PolyMet project.
7 That results in a 104 grams reduction in mercury
8 loading to the St. Louis River over the 20-year
9 life of the mine.

10 Now, a 5.2 g/yr or 104 grams removed over the
11 20-year mine life may not sound like a whole lot of
12 mercury, but when we're talking nanograms per
13 liter, which is a billion times less than a gram,
14 this is a pretty significant reduction in mercury
15 loading to the St. Louis River due to the clean up
16 of brownfield site at this project.

17 We're now going to look at sulfate. As
18 Dr. Branfireun mentioned yesterday, sulfate is also
19 one of the constituents that drives methylmercury
20 behavior which is what everybody is concerned about
21 when we talk about mercury. And it's basically the
22 same story. Sulfate comes in via atmospheric
23 precipitation into the watershed. Some of that
24 makes its way into the wetland systems, into the
25 river, and that's what drives the methylation

1 behavior that we see in these wetland systems,
2 creating the more mobile form of mercury that
3 people are concerned about.

4 So same type of data. We're looking at
5 sulfate loading to these various watersheds. In
6 the upper Partridge River we've got 149 tons per
7 year of sulfate that naturally comes onto the
8 landscape from precipitation.

9 The Embarrass River, 179 tons per year.

10 West Two River, Mud Hen Creek, Sand Creek, and
11 the headwaters to the St. Louis River, 719 tons of
12 sulfate is deposited on the landscape into some of
13 the wetlands. That's what makes its way into the
14 water resources in this system and drives the
15 behavior of methylmercury.

16 740 tons at West Swan River, Flood River,
17 Artichoke River, East Savanna River. And another
18 576 tons per year are coming into the system from
19 Whiteface River and the Flood River.

20 So in total there's about 2,400 tons per year
21 of sulfate that is deposited on this watershed
22 upstream of the Fond du Lac Reservation.

23 When the project goes into operation, again,
24 we're going to be collecting water at the tailings
25 basin through the seepage collection system. All

1 of what I refer to as the contact water at the mine
2 site, that's the water coming out of the mine, the
3 water coming off of the haul roads, et cetera,
4 anything that's going to be carrying sulfate or
5 constituents of concern, all of that water is
6 routed to the water treatment system, and it's
7 treated to very low concentrations, below 1.3 ng/L
8 mercury to 10 mg/L of sulfate before that water is
9 returned to the environment. So we're pulling, if
10 you will, constituents out of the system and
11 returning cleaner water back to the environment.

12 That results in a net effect from the
13 operation where there's going to be
14 1,380,000 kilograms of sulfate that are pulled out
15 of this system every year due to the water
16 management strategy. It's a significant reduction
17 over the life of the project. That is
18 27.6 billion kilograms of sulfate that are going to
19 be pulled out of the system due to the operation of
20 the water management and the water treatment
21 features at this project.

22 I'd next like to turn our attention to
23 specific conductance and salinity. Start out by
24 saying that the project we did provide a memorandum
25 looking specifically at specific conductance. And

1 that memorandum that we provided to the Corps shows
2 that the project will comply with the Band's water
3 quality standard for specific conductance. The
4 project will cause a reduction in specific
5 conductance in the St. Louis River. The project
6 will also cause a reduction in salinity in the
7 St. Louis River at Forbes. We looked at Forbes
8 because Forbes is the furthest upstream spawning
9 area for sturgeon.

10 So the Bands have established water quality
11 standards of 300 $\mu\text{S}/\text{cm}$. Just to clarify, specific
12 conductance refers to the ability of water to
13 conduct electricity. It's based on dissolved
14 anions and cations in the water. Cations like
15 magnesium, calcium and anions like sulfate and
16 chloride and things like that. That's what drives
17 the ability of the water to conduct.

18 So the baseline in the St. Louis River near
19 Cloquet is about 189 $\mu\text{S}/\text{cm}$ at peak project
20 operation. The analysis that we provided to the
21 Corps in our memorandum shows that there will be a
22 0.4 to 0.66 $\mu\text{S}/\text{cm}$ reduction in specific conductance
23 in the St. Louis River. This, again, is due to the
24 fact that we're collecting water coming out of
25 the -- seepage water coming out of the tailings

1 basin that's high in sulfate, high in specific
2 conductance that's going through a membrane
3 filtration system to remove that load of anions and
4 cations in that water. That's going to reduce the
5 specific conductance in the water when that water
6 is returned back to the environment.

7 Likewise, if we're going to reduce specific
8 conductance in the water, we will also reduce
9 salinity.

10 The Band has noted that a salinity of 23,000
11 or 23 parts per thousand will impede sturgeon
12 spawning. We looked at the incremental effect on
13 salinity from the project and there will actually
14 be a reduction in salinity in the St. Louis River
15 at Forbes of between .0007 and .0012 parts per
16 thousand. There will be no impact on spawning of
17 sturgeon due to the operation of this project.

18 Next topic is methylmercury. First, it's
19 noted that mercury methylation will be inhibited by
20 a reduction in sulfate loading from the project.
21 We're pulling a significant amount of sulfate out
22 of the system. That's sulfate that's naturally
23 making its way into the environment today. By
24 pulling that sulfate out of the headwaters that
25 drain into the Embarrass River and the Partridge

1 River, that will reduce the sulfate loading further
2 downstream and will reduce methylmercury formation.

3 We're also going to talk about natural
4 seasonal fluctuations in water levels in these
5 peatland environments as they are the primary
6 driver for mercury methylation, not drawdown.
7 We're going to get into this a little later with
8 some of the subsequent speakers here.

9 First of all, we note that reduction in
10 sulfate loading from the water treatment system
11 will inhibit methylmercury formation. The Band's
12 allegations of increased mercury methylation are
13 predicated on an increase in sulfate loading.
14 There will be no increase in sulfate loading at the
15 project because all that water's going to be pulled
16 out and treated.

17 As noted previously, project-related
18 activities will reduce sulfate loading to the
19 St. Louis River by 1,380,000 kilograms per year
20 over the life of the project. That's
21 27.6 billion kilograms removed from the system.

22 Let's look at the effects of precipitation
23 which brings in mercury, brings in sulfate. Let's
24 look at the effects of precipitation on sulfide
25 oxidation, methylation of mercury, kind of the

1 natural cycle that generates this.

2 So here we have a cross-section schematic of a
3 peat wetland. We've got a shallow water table
4 that's maybe 18 inches, 20 inches below the land
5 surface, and we've got roughly unsaturated,
6 partially saturated peat environment above that.

7 Precipitation impinges on the wetland. And
8 that precipitation, as we discussed earlier, brings
9 in sulfate from the precip and mercury. That
10 sulfate and mercury is then dissolved or available
11 in the pore water of that peat environment.

12 There's also soil dust that is deposited on
13 the landscape in these peat environments, and that
14 soil dust can have sulfide mineralization, sulfide
15 particles.

16 In this upper unsaturated portion of the
17 wetland those sulfide particles can oxidize. And
18 when they oxidize from sulfide, they go to sulfate
19 which is soluble in the water and makes its way as
20 a source of sulfate loading into this peat
21 environment. Over time those particles settle
22 through the peat environment down below the water
23 table. Once they're below the water table, their
24 source of oxygen is cut off and that oxidation no
25 longer occurs, so they're no longer a source of

1 sulfate to the system.

2 As the water table fluctuates and they -- the
3 water levels in these wetlands fluctuate due to
4 seasonal variations and input. So that could be
5 snow melt, variable precip events. Large storms
6 are going to bring the water level up. The snow
7 melt, when it melts, it's going to bring the water
8 level up in the wetland. So when that water level
9 comes up, that sulfate and mercury is mixed into
10 the water table. So that water table goes down.
11 That sulfate and mercury is in an anaerobic
12 environment. And in an anaerobic environment that
13 sulfate, which is now in the groundwater, if you
14 will, is available to bacteria. These are
15 sulfate-reducing bacteria. They respire just like
16 we do. But instead of using oxygen in the
17 respiration process, they are using sulfate. And
18 when they reduce sulfate to sulfide, there's a
19 co-metabolic process that also methylates the
20 mercury. And that's where the mobile form of
21 mercury is generated is through the
22 sulfate-reducing bacteria. That methylmercury is
23 now available to be mobilized in these wetland
24 systems where it can migrate out into adjacent
25 streams and things like that that flow into the

1 St. Louis River.

2 This cycle goes on naturally every day, every
3 year seasonally due to these variable inputs of
4 sulfate, mercury, variable water levels in these
5 wetlands that drives the formation of this
6 compound.

7 It's noted here that the water level
8 fluctuations in these wetlands, which there's been
9 some debate on, are actually -- PolyMet's got data
10 that shows that these water levels fluctuate by
11 about 18 inches per year around the mine site. So
12 we know that this process is naturally occurring
13 and will continue to occur once the project is in
14 operation.

15 So in summary, the analysis of potential
16 effects on water quality, notably mercury,
17 methylmercury, sulfate, and specific conductance as
18 documented in the materials that PolyMet has
19 provided to the PCA and the Corps of Engineers as
20 part of the 401 water quality certification, the
21 cross-media analysis as it's been referred to, was
22 a thoroughly quantitative and exhaustive scope and
23 evaluation to look specifically at the Band's
24 claims that their water quality standards would be
25 0ed.

1 I've been working in this industry for about
2 30 years. I've seen a lot of this type of work on
3 various types of projects. This was very
4 exhaustive and it was very much on point directly
5 to the questions that the Bands have raised
6 throughout the EIS and throughout the permitting
7 process.

8 The analysis, which is conservative, shows
9 that the project will reduce loading of sulfate and
10 mercury to the St. Louis River. There will be no
11 violation of the Band's water quality standards for
12 specific conductance, sulfate, or any other
13 standard.

14 There will be a reduction in specific
15 conductance in the St. Louis River. There will be
16 a reduction in salinity in the St. Louis River.
17 There will be no impact on sturgeon spawning. And
18 water level fluctuations in the wetlands will not
19 alter the generation of methylmercury as has been
20 alleged in prior presentations.

21 So with that, I'll hand it over to the next
22 speaker.

23 MR. CLIFF TWAROSKI: Hello. My name
24 is Cliff Twaroski, T-W-A-R-O-S-K-I. I am an
25 environmental scientist with Barr Engineering

1 Company. I've been with Barr for almost 25 years.
2 Prior to that, I was with the Minnesota Pollution
3 Control Agency for a little over 15 years. And my
4 thesis work was done on the Krone bog just to the
5 west of the casino here looking at peatland
6 reclamation.

7 Today I'd like to talk about points of the
8 project as far as the detailed modeling that shows
9 we have a reduction of mercury and sulfate as well
10 as methylmercury.

11 Unfortunately, I do need to talk about the
12 Band not accounting for the project water
13 management and treatment. It's an important part
14 of the project that needs to be accounted for.

15 We also have heard the Band and their concerns
16 about flushing events. And so we have done a
17 screening analysis to address that. And I'll talk
18 more about that. But that screening analysis is
19 indicating that the project is still reducing
20 sulfate, mercury, and methylmercury loading even
21 under those high flushing events.

22 We heard a little bit about linkages
23 yesterday. And I'd just like to refresh
24 everybody's memory for methylmercury that is linked
25 to sulfate and mercury and the anaerobic

1 environments and wetlands and lake sediments. For
2 the export of mercury and methylmercury from
3 wetlands, that's linked to organic matter and water
4 flow. Organic matter is a carrier of mercury and
5 methylmercury. And as increasing flows occur, you
6 will be increasing organic matter export. The DNR
7 has documented that and there's studies of the
8 St. Louis River and we'll talk a little bit more
9 about that later.

10 Fish uptake of methylmercury is linked to the
11 formation of methylmercury which is linked to the
12 sulfate and mercury. And the export from wetlands
13 to downstream waters is linked to the water flow.

14 And the main point here is that if we affect
15 one part or one parameter, we are going to be
16 affecting other parameters as well.

17 I'd like to talk a little bit more about the
18 sulfate and methylmercury linkage with mercury
19 being transformed to methylmercury, again,
20 occurring primarily in wetlands and lake sediments.
21 The methylation process does not so much occur in
22 flowing waters where you have more oxygen occurring
23 and we call that channel flow. Methylmercury, once
24 it's in the food chain, as Dr. Branfireun pointed
25 out yesterday, there is biomagnification up the

1 food chain, and that has resulted in a number of
2 fish consumption advisories, including for the
3 St. Louis River.

4 One of the other points I'd like to make about
5 this sulfate methylmercury is that MPCA did an
6 analysis around a tailings basin. And they found
7 that if you increase sulfate load, you will be
8 increasing methylmercury.

9 The amount of sulfate increase is important,
10 though. In this watershed, in the St. Louis River
11 watershed an increase in sulfate may not show an
12 increase in methylmercury and that is included in
13 the DNR studies as well as PolyMet studies.

14 The MPCA has also concluded in their statewide
15 mercury TMDL that sulfate loading will -- decrease
16 in sulfate loading will decrease mercury in fish.
17 USEPA approved that TMDL. And by that approval,
18 they also concur with that finding. Also, the
19 USEPA in another report, I think in 2009, also
20 identified that if you reduce sulfate loading, you
21 are reducing methylmercury.

22 We'd also like to talk a little bit about
23 existing conditions. And we are talking about this
24 to make sure that everybody has a proper
25 understanding of how the project fits into these

1 existing conditions. This is sulfate. And
2 sulfate, as you heard yesterday, there's sulfate
3 coming from the mining watersheds which are these
4 dark gray watersheds on the top part of the
5 St. Louis River watershed. And with the loading
6 estimates, the mining watersheds provide about
7 70 percent of the sulfate load in the St. Louis
8 River.

9 You can see that from the proposed project
10 area the former LTV tailings basin provides some
11 sulfate as well. There's also sulfate provided by
12 the future mine site. Even though it's not -- it's
13 a natural site right now, there is sulfate loading
14 that is occurring from that area.

15 The other part of the slide is that there is
16 sulfate coming from non-mining watersheds. And as
17 the DNR has identified, that is a significant
18 loading to the lower St. Louis River. If you took
19 away all the mining sulfate, you would still have a
20 methylmercury problem in the lower St. Louis River.
21 And during storm events, high flow events like
22 occurring now, the sulfate loading from the
23 non-mining watersheds, in particular these large
24 watersheds on the southern part of the St. Louis
25 River watershed, can contribute as much sulfate as

1 the mining watersheds. So they are not
2 insignificant. And we just want to make sure
3 everybody understands that.

4 For methylmercury and for mercury, the
5 contribution is primarily from the non-mining
6 watersheds. And again, this is different than
7 sulfate.

8 Here, the non-mining watersheds, particularly
9 the Cloquet River watershed and the Whiteface River
10 watershed, are primary contributors of
11 methylmercury loading to the lower St. Louis River.
12 Both of these systems come in below the mining
13 district. And the Whiteface comes in just north of
14 the Fond du Lac Reservation and Cloquet comes in
15 within the Fond du Lac Reservation boundaries.

16 I did not hear the Band identify this type of
17 loading to the lower St. Louis River yesterday nor
18 have I found any of this loading information in any
19 of their documents that they've presented or
20 prepared as part of the environmental review
21 process or in this proceeding.

22 Again, the non-mining areas are the major
23 contributor. One of the reasons for the Whiteface,
24 that watershed has about 31 percent of its area in
25 wetlands and that is contributing to its

1 contribution.

2 The Cloquet River watershed, though, only has
3 about 13 percent of its watershed in wetlands. But
4 the Cloquet also has a very high flow. It's a
5 major contributor to flow in the lower St. Louis
6 River. So even at moderate methylmercury
7 concentrations, its higher flow produces a loading
8 that is just slightly less than the Whiteface
9 River.

10 The one interesting point on this slide is
11 that for the Partridge River it's identified as a
12 mining watershed, and it's standing out a little
13 bit more than, say, the Embarrass River watershed
14 or the Swan. Those are also mining watersheds.
15 And for the Partridge there is not much mining
16 development in that watershed. It has the north
17 shore Peter-Mitchell pit dewatering water that is
18 provided to the Partridge. But otherwise, that
19 watershed is fairly undisturbed.

20 But one of the things about the Partridge is
21 that it has two major sub-watersheds within it, the
22 south branch of the Partridge as well as Colville
23 Creek. And those two tributaries to the Partridge
24 enter the main stem of the Partridge below the
25 project area, but they are originating wetlands.

1 And that baseline data we have for the Partridge
2 River indicates that they provide more loading to
3 the Partridge River than does the upper part where
4 the mine, proposed mine area is located.

5 So again, wetlands within the Partridge River
6 are providing this methylmercury load. And the
7 project will have no effect on 99 plus percent of
8 the loading of methylmercury to the St. Louis
9 River. The project will address -- will be
10 reducing loading from its area, but overall it will
11 not have an affect on the majority of methylmercury
12 showing up in the lower St. Louis River.

13 Now let's talk about the project. And the
14 project does have additions. We do have a
15 wastewater treatment discharge. We have sulfate in
16 that discharge at 10 mg/L. We have mercury in that
17 discharge at 1.3 μ S/L. And it does add up to a
18 load going out to those wetlands. And the Band has
19 focused on this part of the project.

20 However, if you really want to look at the
21 overall project, and you need to look at the
22 overall project, there is a lot of reductions in
23 sulfate and mercury occurring with the project in
24 operation. And when we look at, in particular, the
25 headwater wetlands, which the Band has identified

1 as an important area of concern, the loading of
2 sulfate is reduced by some 265,000 kilograms per
3 year. And if you multiply that number by 20, it
4 gets up very high over the life of the project.
5 Mercury is also reduced to those headwater wetlands
6 about 2 grams per year.

7 And the other part that I'd like to emphasize
8 is that with the wastewater treatment system
9 discharge, that discharge needs to be within plus
10 or minus 20 percent of existing conditions average
11 annual flow, and that is what is showing up on this
12 slide. And so with the load reductions to the
13 headwater wetlands due to the water capture and
14 treatment, there is no increase in loading to those
15 wetlands.

16 I'd like to talk about that water and water
17 loading a little bit more because that seems to be
18 an important part of the Band's comments.

19 Again, the wastewater treatment discharge
20 needs to be within plus or minus 20 percent of the
21 average, annual average existing conditions flow.
22 The Band claimed that there will be excess flushing
23 of these wetlands. That is not -- that can't
24 happen if we're within plus or minus 20 percent of
25 existing conditions.

1 In addition, the Band has said that the
2 flushing of the wetlands will increase organic
3 matter export downstream. If we're not increasing
4 flows, not having excessive amounts of water being
5 released to those headwater wetlands, then we will
6 not be flushing more organic matter that will carry
7 more mercury and more methylmercury downstream than
8 what is already occurring.

9 So again, we're staying within pretty much
10 existing conditions with the wastewater treatment
11 system discharge to those headwater wetlands, and
12 water loading is not an issue for this project.

13 For the sulfate for project impacts we looked
14 at a number of evaluation points around the project
15 in the Embarrass River, the Partridge River
16 watershed, as well as locations in the St. Louis
17 River and the Partridge River. Or sorry. In the
18 St. Louis River upstream of the Fond du Lac
19 Reservation as well as downstream.

20 For the plant site area in the Embarrass
21 River, we found, again for the headwater wetlands,
22 that we will have reductions of sulfate, about
23 126,000 kilograms per year, to the Trimble Creek
24 headwater wetlands, about 139,000 kilograms per
25 year to the unnamed creek headwater wetlands, and

1 by the time we get down to PM13 in the Embarrass
2 River itself, we have a reduction of about
3 1,280,000 kilograms per year. And this is
4 cumulative loading. This accounts for the
5 project's air emissions of sulfur as well, and that
6 includes sulfur emissions from stacks, mobile
7 sources, as well as fugitive sulfide mineral dust.

8 For the mine site we also see a reduction of
9 sulfate, about 9,000 kilograms per year, due to
10 water capture and treatment. We also see a
11 reduction at discharge location SD026 which is the
12 headwaters of Second Creek. That reduction is a
13 little more than 84,000 kilograms per year. And
14 again, that is wastewater discharge at about
15 10 mg/L.

16 We also see a reduction of about
17 15,000 kilograms per year due to the transfer of
18 Colby Lake water to the plant site for use as
19 process water. And overall, when we look at the
20 Partridge River, we have about 100,000 kilograms
21 per year reduction. And again, this is taking into
22 account the project's air emissions. In the
23 Partridge River most of those air emissions are
24 being driven by the sulfide mineral dust, and
25 that's accounted for in these loading reductions.

1 When we get to the St. Louis River, when we
2 combine up the reductions in the Embarrass River
3 and the Partridge River, we're looking at a
4 reduction of about 1,380,000 kilograms per year.
5 And again, when we look at 20-year life, we're in
6 that approximately 28 billion kilograms per year of
7 sulfate reduction that's getting to the river.

8 Forbes is above the Fond du Lac Reservation.
9 This loading reduction carries down to Cloquet and
10 to the Band's boundaries.

11 If we are reducing sulfate loading in the
12 headwaters, we cannot be increasing sulfate loading
13 downstream. If we're reducing sulfate loading
14 around the plant site, mine site, that means we are
15 also reducing methylmercury both around -- in both
16 the Embarrass and Partridge River watersheds as
17 well as downstream at the Fond du Lac Reservation.

18 For mercury we also evaluated loading close to
19 the project in the same locations as we did for
20 sulfate and also in the St. Louis River. For that
21 analysis the numbers are smaller but we are still
22 showing a reduction of mercury loading to those
23 headwater wetlands near the tailings basin, 1.5
24 gram per year reduction to Trimble Creek.

25 Headwater wetlands, a small reduction, about

1 0.2 grams per year to the unnamed creek headwater
2 wetlands. And when we get out to PM13 in the
3 Embarrass River itself, the reduction is about
4 0.8 grams per year. And again, this includes the
5 project's air emissions of mercury which were
6 modeled out to about 10 kilometers away from the
7 project and estimating deposition from those
8 emissions.

9 At the mine site we see a reduction of about
10 6 grams per year. We also see a reduction when we
11 transfer Colby Lake water over to the plant site.
12 However, we also see an increase at SDO26 which,
13 again, is the headwaters of the Second Creek. That
14 increase is due to a small increase in water flow
15 due to the wastewater treatment system. And it's
16 also an increase based on our assumption that the
17 mercury concentration of 0.5 ng/L now goes up to
18 1.3 ng/L with the wastewater treatment discharge.
19 And that was a conservative assumption that we made
20 for our impact calculations.

21 However, as you've heard two previous speakers
22 talk about, the Eagle Mine in Michigan has the same
23 technology for water treatment as being proposed
24 for the PolyMet project. My understanding is
25 there's at least three years of data that are

1 showing mercury concentrations in that discharge
2 water of about 0.5 ng/L or less. So the increase
3 that we have identified here for SD026 is likely an
4 overestimate and that with the project in actual
5 operation, any increase in mercury is likely to be
6 less than what we have calculated.

7 So when we look at the reduction of mercury
8 loading in the Embarrass River of .8 grams per
9 year, a reduction from the Partridge River
10 watershed of 4.4 grams per year, that totals to a
11 little over 5 grams per year reduction in the
12 St. Louis River at Forbes. And again, Forbes is
13 upstream of Fond du Lac Reservation. This
14 reduction carries down to the Fond du Lac
15 Reservation as well as to Cloquet. And because of
16 the relationship of mercury, sulfate, and
17 methylmercury, if we're reducing mercury in the
18 headwater regions, then we will be reducing
19 methylmercury in the lower St. Louis River as well.

20 So we've conducted a number of evaluations for
21 the project. We've looked at the Embarrass River,
22 Partridge River, St. Louis River at Forbes and
23 Cloquet, and all of those are identifying that
24 there will be a decrease in sulfate and mercury
25 loading and methylmercury loading to the watersheds

1 where the project is located as well as to the
2 St. Louis River at Forbes and Cloquet.

3 Cross-media analysis was conducted to
4 specifically address the Band's concerns about
5 sulfide mineral dust adding sulfate to wetlands and
6 the methylmercury formation within those wetlands
7 due to this extra loading of dust.

8 What you can see in this table is that on an
9 overall basis, the loading of sulfate from the
10 project is small. And the same for mercury.

11 And when we look at the historic loading,
12 especially of sulfate, which has been much higher
13 than it currently is, and we take into account that
14 methylation has occurred at higher background
15 loading, it's occurring now under existing
16 conditions, and this small potential increase from
17 the project, our conclusion is that we are not
18 changing the methylating environment of wetlands in
19 and around the project.

20 But the Band still has concerns, and they're
21 expressing concerns about flushing events. And so
22 we took a look at those flushing events to see what
23 it meant. And in particular, the methylmercury was
24 of interest to us because -- and as it is to the
25 Band as to what's happening when we have these

1 flushing events.

2 The one thing that we are doing here is that
3 we are comparing a flushing event in existing
4 conditions to a flushing event with the project in
5 operation. All of our previous mass balance
6 calculations and analyses have been on average
7 conditions. And the Band has seemed to take those
8 average conditions and compared them to flushing
9 events. And under that kind of comparison, yes,
10 there is an increase -- the project would show an
11 increase. But we want to look at a fair comparison
12 of the project in operation to a flushing event in
13 existing conditions.

14 So we assimilated data for looking at maximum
15 flows from the project's water modeling data
16 packages. We looked at mercury and methylmercury
17 concentrations from the baseline data that's been
18 collected for the project, as well as supplemental
19 information from the DNR studies and Pollution
20 Control Agency studies.

21 For the Embarrass River and Partridge River
22 watersheds in particular, we're using a baseline
23 data but there is some DNR data for both of those
24 watersheds.

25 For the St. Louis River at Forbes and Cloquet,

1 that data is primarily coming from DNR reports,
2 USGS flow information, as well as the Pollution
3 Control Agency and some of their data.

4 And again, the project in operation does
5 reflect water capture and treatment. And that,
6 again, is a primary part of the project that needs
7 to be incorporated into all of these understandings
8 of what this project means.

9 So when we look at the project and a high
10 flushing condition, we can see that there is still
11 a reduction occurring even under these flow
12 conditions, again, because of the water treatment
13 and water capture occurring due to the project.

14 The other part of this information is that
15 this net change, this reduction due to the project
16 carries down to Forbes, but we also see a very high
17 contribution from non-project areas. And as you
18 recall some of the slides presented by Steve
19 Donohue, there is a significant amount of mercury
20 that's being deposited to the St. Louis River
21 watershed as a whole, and that's being reflected in
22 these non-mining area contributions, non-project
23 area contributions as being much more significant
24 than what's coming out of the project area in
25 existing conditions. And that's the same whether

1 you're at Forbes or whether you're at Cloquet. And
2 again, we have the Whiteface River and Cloquet
3 River coming in in this lower part of the river,
4 and they are major contributors to this non-project
5 area loading.

6 With the project in operation, there is,
7 again, a reduction in mercury loading. And that is
8 again carried through to Forbes and it's also
9 carried down river to Cloquet.

10 For methylmercury we have the same story that
11 the project does reduce loading of methylmercury
12 under these high flow scenarios. Again, that
13 loading is occurring at Forbes up above Fond du
14 Lac. It's also occurring downstream at Cloquet.
15 And again, we have a large contribution from the
16 non-mining, non-project areas and only a small
17 contribution from the project area itself whether
18 it's existing conditions or in operation.

19 And again, the overall conclusion here is that
20 we have, under this worst-case scenario where --
21 and this incorporates all the severe flushing,
22 severe water level declines that the Band has
23 talked about, this incorporates all of that and
24 still we show a reduction in both mercury and
25 methylmercury under this high-flushing scenario.

1 So in summary, our modeling work and the
2 support analyses that we've conducted identify
3 decrease in sulfate, mercury, and methylmercury
4 loading in the St. Louis River.

5 The Band does not seem to account for this
6 water management treatment. The direct discharges
7 from the project will not increase loading of
8 water, organic matter, sulfate, mercury, or
9 methylmercury.

10 The wastewater treatment discharge to
11 headwater wetlands will be similar to existing
12 conditions flow, so there's, again, no excess water
13 loading, no excess flushing of organic matter.

14 If we are reducing loading in the headwaters
15 region, we will not be increasing loading in
16 downstream areas. And even under a high-flushing
17 scenario, the project is still reducing loading.
18 And again, if we're reducing sulfate in the
19 headwaters region, we are reducing methylmercury in
20 the downstream waters.

21 And so that's the end of my preparation.

22 MR. GREG COUNCIL: Good morning. My
23 name is Greg Council. That's spelled
24 C-O-U-N-C-I-L. I am an environmental engineer with
25 Tetra Tech.

1 My background is I have about 28 years of
2 experience focused on groundwater hydrology,
3 groundwater modeling, and the interaction of
4 groundwater and surface water and modeling of that
5 process, those processes. My work on the project
6 to date primarily has been peer review-type work in
7 the modeling area.

8 Today I'm going to discuss the claims that
9 were made by the Band related primarily to
10 groundwater drawdown and how such a groundwater
11 drawdown might affect water quality in downstream
12 waters.

13 The items I'll be talking about, the analyses
14 are documented in a memo that we provided to the
15 Corps.

16 As Ms. Schuldt did yesterday, I'd like to
17 start with a map of the watershed just to orient
18 ourselves. You've seen this map before. I'll
19 point out that for the purpose of the presentation
20 today, we're going to focus right here just on the
21 mine site where drawdown from development of mine
22 pits would occur because of the groundwater that
23 would flow into those pits.

24 So focusing now just on that area, let's look
25 at these features around in the mine site.

1 This particular figure shows the Partridge
2 River flowing around the mine site. This is the
3 flows on the north side, the east side, and then
4 the south of where the mine pits would be. Those
5 mine pits are outlined here in black, the west pit
6 and the combined central and the east pit. Not
7 developed yet but if the mine is permitted, these
8 pits would then be excavated and groundwater would
9 flow into them.

10 Shaded in green here are the extensive
11 wetlands throughout this watershed. We've talked a
12 lot about the wetlands in the mine site area. And
13 we're going to talk a little bit about them over
14 the next few minutes. I'll just point out that,
15 yes, there are extensive wetlands in this area. It
16 covers a lot of this map.

17 I'll also point out that in the upper
18 left-hand corner you see a portion of the
19 Peter-Mitchell mine, an active, existing iron mine
20 that's somewhat near the proposed mine site.

21 So I want to point out, just make sure that
22 we're all clear and we all agree, I believe, that
23 these wetlands do now produce sulfate, mercury,
24 methylmercury. These are constituents that are
25 currently stored in the peats of these wetland

1 sediments and can be released when sulfur is
2 oxidized and then that sulfur promotes
3 methylmercury creation from the mercury through the
4 sulfate-producing bacteria. This is happening now.
5 It's happening in this watershed and all throughout
6 the St. Louis River watershed, as Cliff Twaroski
7 pointed out earlier.

8 So we're going to talk a little bit about the
9 claim that there would be a massive drawdown here
10 as a result of development of these pits and that
11 that massive drawdown, according to the Band, would
12 lead to drastically more sulfate oxidation and that
13 sulfate oxidation in the sediments would lead to
14 methylmercury creation that would then be
15 transported to the Partridge River and then
16 downstream to the St. Louis River.

17 We're going to talk about that and try to at
18 least give some sense of the quantities involved
19 here. There's really not a quantification of those
20 processes in the Band's claim but we'll try to at
21 least provide some calculations in this discussion.

22 So to overview the Band's claims here related
23 to drawdown. The Band -- and also, a little bit of
24 my summary that is detailed more in the memo.

25 Basically we're going to show in some of these

1 subsequent slides that the Band is really not
2 accounting for the fact that wetlands in this area
3 will actually be directly removed. And while that
4 will have to be mitigated and is, in fact, being
5 mitigated, the removal of the wetlands actually
6 removes the sulfate-generating portion of those
7 wetlands.

8 The Band's analysis overstates the amount of
9 drawdown that would occur. We'll get into that a
10 little bit. And through that overstatement,
11 basically implies that you're going to get a net
12 increase of sulfate, mercury, methylmercury
13 reporting to the Partridge River and to downstream
14 waters. We'll go through those claims and show you
15 why that is not the case.

16 Additionally, the Band claims that MODFLOW
17 should have been used to predict what the impacts
18 would be in wetlands. Going to talk about that on
19 one slide just to show that that's really not what
20 MODFLOW is intended for.

21 And then lastly, we'll talk about some
22 mitigating factors, some hydrogeochemical
23 mitigating factors that influence what actually
24 happens with the sulfate and the methylmercury so
25 that we get a little bit better handle on what's

1 really going on in the system.

2 Back to this figure. Again, this is just an
3 overview figure showing the area around the mine
4 site. I'll superimpose on that the acreage of
5 wetlands that would be removed, basically filled
6 and excavated, as a product of building these mines
7 and the stockpiles. These wetlands now, once the
8 project would go into effect, no longer generate
9 sulfate or methylmercury.

10 Going one step further, this figure shows some
11 of the wetlands that would be potentially impacted
12 by drawdown. This is from the analysis in the
13 FEIS, so this analysis is based on the analog
14 method which uses the data from a nearby mine to
15 estimate how much drawdown would occur in and
16 around the mine. We all agree that drawdown would
17 be greater near the mine and near the mine pits and
18 that it would be decreased as you move away.

19 This one, this particular rendition shows the
20 drawdown -- or shows in orange, I should say, the
21 areas that are in the FEIS predicted to be highly
22 likely affected by drawdown or moderately likely
23 affected by drawdown. The FEIS also points out
24 some wetlands that are low likelihood drawdown
25 wetlands.

1 But this acreage, high and moderate, is
2 160 acres, a little bit more than 160 acres.
3 Again, it's based on the analog method from the
4 nearby Canisteo Mine. It's conservative in one way
5 in that the Canisteo Mine is actually developed
6 into the Biwabik formation, which is a permeable
7 relative to the Duluth Complex and the Virginia
8 formation where these mine pits, where the PolyMet
9 mine pits would be developed. The Canisteo Mine is
10 in a much more permeable unit.

11 So if we look at these orange-shaded wetlands
12 that are potentially affected by drawdown, they
13 obviously -- the obvious question is what if the
14 high -- what if these drawdowns were affected by
15 drawdown in a way that increased the oxidation,
16 therefore, increasing sulfate and increasing
17 methylmercury production.

18 So as I said, you've got about -- I may not
19 have mentioned the acreage. You've got about
20 750 acres taken away as methylmercury producing
21 acreage of wetlands, so those now produce sulfate,
22 methylmercury, would be taken way from this system
23 producing sulfate methylmercury. And you still
24 got -- maybe you've got 160 or so acres that have
25 an additional oxidation capacity because there's

1 drawdown that creates a little bit dryer wetland.

2 Well, in order to make up just for the 750
3 that you've removed, you'd have to more than
4 quintuple the amount of sulfate creation from those
5 161 acres that have been increased in oxidation.
6 Nowhere do any of the data or any of the scientific
7 studies suggest that this type of increase is
8 likely.

9 One of the studies, I believe, that
10 Dr. Branfireun pointed out yesterday showed 190
11 percent increase in methylmercury. But again, not
12 quintupling of that amount.

13 It's important. You can't ignore the loss and
14 load due to the removal of wetlands in capture of
15 the water. The net effect here is a reduction in
16 methylmercury creation.

17 Now, what if the area of impact is much larger
18 as the Bands claim. This figure, I've zoomed out a
19 little bit so that we can show the entirety of
20 what -- approximately at least is what the GLIFWC
21 analysis, the Band's analysis, has alleged would be
22 the actual affected acreage based on their
23 alternative analyses.

24 So this shows in blue, combined with the
25 previously shaded orange and black areas, about

1 6,000 -- actually a little bit less but roughly
2 6,000 acres that are within 10,000 feet of the
3 proposed PolyMet pits.

4 So I wanted to look at this 6,000 acres and
5 say, well, if the drawdown is massive enough to
6 create this big of a drawdown, this much wetland
7 impact, how much would the wetland impact be on
8 average? Or how much -- how could I get a bounding
9 calculation to see what additional loads of sulfate
10 and methylmercury would actually be generated from
11 this much wetland impact.

12 And here, I think it's important first to just
13 sort of -- first of all, we just developed a little
14 illustration to show how big the 6,000 acres is
15 compared to what's being removed and what we really
16 think would be affected by drawdown.

17 So this shows -- this is just a simple
18 illustration. Again, in blue just an area that's
19 6,000 acres if every square on this figure is
20 assumed to be one acre. We'll compare to that the
21 amount of wetlands that would be directly affected,
22 basically removed through filling or excavation.
23 That's about 750 acres. And we'll also show the
24 area that the FEIS shows would be highly or
25 moderately likely to be impacted by drawdown, a

1 much smaller 161 acres. But if it was -- if
2 drawdown actually affected the entirety of the
3 6,000 acres, how severe would that impact be?

4 For this it's important to look at how much
5 groundwater is actually going to be flowing into
6 the pit. We have estimates of this from the detail
7 modeling, the MODFLOW modeling that was developed
8 for the project.

9 The MODFLOW, by the way, the groundwater
10 model, is a good tool for estimating groundwater
11 inflow to a mine pit.

12 So this figure shows, on the X axis, the year
13 of operation of the mine. And the Y axis shows the
14 estimated modeled inflow to the groundwater pits
15 from groundwater. Each pit, the central pit, the
16 east pit, and the west pit, are shown on different
17 line, different colors with the black line being
18 the total groundwater inflow to the entire mine.

19 It averages over time, if you just take a
20 simple time average over this entire 20-year mine
21 life, the average inflow is about 502 gallons per
22 minute. Roughly 1.1 CFS, just for context. I
23 think Christie Kearney pointed out earlier that the
24 flow in the Partridge River just downstream of the
25 mine is about 49 CFS on average, so we're talking 2

1 to 3 percent of the total flow is groundwater flow
2 coming into the mine.

3 Where is this groundwater flow coming from?

4 Well, it's not all coming from just wetlands but it
5 could be -- some of it could be coming from
6 wetlands. A lot of it's coming from groundwater
7 storage because as you draw down the water table,
8 you're pulling water that had previously been
9 stored in the pores and the unsaturated -- and the
10 unconsolidated system, and you're pulling in water
11 that was previously in fractures. You're also
12 perhaps pulling in water that would have been
13 evapotranspired in uplands. And you are pulling
14 in water that would have reported instead to
15 wetlands or maybe even more that seeps out of
16 wetlands.

17 So let's assume that a lot of this water,
18 maybe even all of this 500 gallons a minute, was
19 pulled from these 6,000 acres of wetlands. What
20 would that really mean?

21 So if we take the Band's assertion that 6,000
22 acres would be affected, on average, pulling out
23 500 gallons a minute out of 6,000 acres of wetland
24 results in a .083 gallons per minute per acre
25 removal out of those wetlands.

1 Now, we've made some assumptions here,
2 conservative for the most part in that we've taken
3 the entirety of the predicted groundwater inflow
4 and assumed it's all taken from the water budgets
5 from these 6,000 acres of wetlands around the mine,
6 within 10,000 feet of the mine.

7 So with that average effect, we're pulling out
8 basically 1.6 inches per year out of the water
9 budget of the wetlands around the mine in this
10 particular analysis. That 1.6 inches per year is
11 about 5 percent of the average precipitation in
12 this area of Minnesota. Average precipitation
13 being around 30 inches per year.

14 So we're pulling out perhaps 5 percent of the
15 water budget of these wetlands through development
16 of the mine, pulling groundwater into the mine
17 instead of going to wetlands. We basically, in a
18 way, made the wetlands 5 percent dryer. If we take
19 what that 5 percent dryer wetland really means,
20 it's effectively like having the original 6,000
21 acres and adding another 5 percent. So you've got
22 6,000 acres that are already now producing sulfate
23 through these processes, natural variation in water
24 levels going up and down. You get sulfate
25 creation. You get methylmercury creation. If you

1 think about a 5 percent dryer case, it's
2 effectively like adding 300 acres back in. That's
3 5 percent of 6,000. You still haven't accounted
4 for the removal -- you still haven't made up for,
5 rather, the removal of the 750 acres through the
6 original direct removal of wetlands.

7 So you still, in this analysis, have a net
8 loss of sulfate, methylmercury, and mercury
9 reporting to the pore waters and the wetland
10 sediments.

11 Let's talk for a second about MODFLOW. As I
12 mentioned, MODFLOW, very good tool for estimating
13 what's going on in groundwater.

14 In this case we've used MODFLOW to calculate
15 mine inflow. That's a good use. There is many
16 limitations with using MODFLOW to predict what's
17 happening in wetlands. It really doesn't have the
18 capability of simulating wetlands in any detail.
19 And that's important because wetlands are complex.
20 They're very variable on a spatial scale and on a
21 temporal scale. There are natural fluctuations in
22 the way that wetlands behave that go on now, and
23 these are hard to capture, especially with models
24 that have a large grid cell size and typically have
25 a long time still.

1 Importantly, wetlands have low permeability
2 peat sediments in many cases. And those, without
3 doing a lot of layering and a lot of detail, real,
4 real fine-tuning, it's just really, really
5 difficult to get that right even if you were to use
6 something like a numerical simulator to try to
7 predict that.

8 There are other wetland processes going on.
9 They're basically impossible to use -- impossible
10 to model with MODFLOW. So while it's a useful
11 tool, the important limitations of MODFLOW with
12 regard to wetlands make it really not useable for
13 predicting directly what's happening in wetlands.

14 Some of the complications actually also come
15 into play when you think about how the sulfate and
16 mercury and wetland sediments may mobilize down
17 gradient.

18 For this I'm going to turn a little bit to
19 just a brief overview of the hydrogeochemical
20 conditions that have to occur for the sulfate to
21 oxidize and to create methylmercury. I won't go
22 into this in detail because we've covered it. We
23 covered it yesterday with Dr. Branfireun. We
24 covered it today on a couple previous
25 presentations. Just to say that we all know it's

1 happening that naturally this happens and that,
2 yes, in certain circumstances it could be
3 exacerbated by drawdown.

4 But there are at least four mitigating factors
5 that we're going to go into that describe why --
6 you take a little bit bigger picture, you take a
7 step back, any sulfate, methylmercury, mercury
8 that's created in the pores of the wetland
9 sediments, instead of reporting down to the
10 Partridge River and downstream waters, is actually
11 going to report to the mine, or otherwise, not go
12 downstream and would eventually be pumped over to
13 the plant site where it would be eventually treated
14 by the reverse osmosis, by the membrane treatment
15 system that we discussed earlier.

16 So in the bigger picture, this
17 cross-section -- it's just a conceptual
18 cross-section -- this cross-section shows a little
19 bit of the processes that actually happen once a
20 mine pit is developed and water begins to drawdown
21 near it.

22 So in much of the area around the mine what
23 you have is a decreased water table and water that
24 would have flowed via runoff out to the Partridge
25 River, would have potentially flowed via shallow

1 groundwater flow. Maybe during high storm events
2 the groundwater would have discharged to the
3 surface and then through runoff. A lot of that
4 water now is going to report to the mine pit
5 instead. Again, it just gets captured and
6 eventually treated. You get less runoff. You get
7 less flow of groundwater, lower gradient. I'll
8 talk about -- basically less driving force of
9 groundwater toward the Partridge River. And you
10 actually get some mobilization of mercury downward
11 into the soil column which tends to sequester it to
12 some degree.

13 So effectively, these processes are mitigating
14 in that they limit the downstream, the down
15 gradient movement of the sulfate and methylmercury
16 to downstream waters.

17 As I said, there are at least four processes
18 here that we'll just briefly touch on.

19 These processes mitigate the transport of the
20 sulfate and mercury to downstream waters.

21 The first one is you've reduced now, through
22 the development of the mine, the hydraulic gradient
23 that naturally goes from the mine site area, the
24 upland mine site area, down toward the Partridge
25 River. You've reduced that hydraulic gradient and

1 you, therefore, reduce the driving force that would
2 push groundwater toward the Partridge River. This
3 results in a lower load of water, of sulfate and
4 other constituents to the Partridge River.

5 In fact, nearest the mine where the
6 drawdown -- where the increased oxidization is
7 alleged to occur, nearest the mine, that is where
8 you're more likely to have the water flowing toward
9 the mine where it would eventually get captured.

10 Second mitigating factor. As I mentioned,
11 you've pulled now the water table down so that
12 during high flow events like the large snow melt
13 event, like really what's going on now with the
14 high flows in the rivers, during these high runoff
15 events, you're less likely to have wetland pore
16 water discharging up into runoff and making it to
17 the Partridge River and then to downstream waters.

18 So during these high events, you're going to
19 have greater infiltration, a greater balance of
20 more infiltration and less discharge, and you're
21 going to have less runoff, less sulfate and
22 methylmercury making it to the rivers.

23 Thirdly, mitigating factors. There will be
24 some vertical redistribution of methylmercury
25 downward into the soil column. Some of the

1 experiments show that -- and these are some of the
2 experiments, I think that Dr. Branfireun was
3 talking about yesterday -- they show that actually
4 the process moves -- the process of the cycling
5 that was illustrated well, I think, in Steve
6 Donohue's animations earlier, that process tends to
7 move the mercury down in the sediment column during
8 the process once there is some drawdown underneath.

9 This graph shows -- this was pulled directly
10 from one of those -- it's footnoted here --
11 directly from one of those papers. It shows the
12 mercury concentrations in soil, in sediments of the
13 wetland, rather, measured as a function of depth.
14 So we have concentration on the X axis up here and
15 we have depth on the Y axis.

16 In the high water table or base case, let's
17 say, the water tables varying between 7 and
18 11 centimeters below land surface. And in this
19 case you've got your peak mercury concentrations
20 occurring at about 20 to 35 centimeters below land
21 surface. Once the low water table is established
22 in the same sediment column, what happens is you
23 get the higher mercury concentrations, similar
24 levels but much lower in the soil column down to 30
25 to even 60 centimeters deep. So you got less

1 mercury in the more available, shallower portion of
2 the soil column and more mercury down deeper.
3 You've effectively sequestered some of the mercury
4 into a deeper portion of the sediment column.

5 Finally, demethylation. Well, we all agree
6 that the methylation of mercury is an important
7 process. It's important to also consider that
8 demethylation occurs. So it's a reversible process
9 and it can occur where the actual mercury that's
10 created, rather than making it downstream to the
11 Partridge River, could be, by the same basic
12 processes, could be demethylated.

13 And in fact, one of the studies that, I
14 believe, Dr. Branfireun talked about yesterday
15 talks about this demethylation and about how
16 important it is in that it prevents what you might
17 expect otherwise to see in the downstream waters.

18 So in summary, we find that the Band's "will
19 affect" analysis related to drawdown and its
20 creation of mercury -- methylmercury, rather,
21 sulfate in downstream waters is overstated not only
22 because it overpredicts the drawdown in wetlands
23 relative to what the FEIS describes, it also
24 implies that they -- that the oxidation in the
25 sediments in these larger area of wetlands would

1 overwhelm the other impacts that we've talked about
2 before that tend to reduce the loads of sulfate,
3 mercury, and methylmercury to downstream waters.

4 It's important to not only assess what might
5 happen due to drawdown to increase oxidation, but
6 also the important mitigating factors that I
7 describe that would tend to pull any increased
8 sulfate and any existing sulfate and mercury and
9 methylmercury into the mine where it would be
10 treated before discharge.

11 So as part of the project, not only do we
12 expect the sulfate and mercury and methylmercury to
13 decrease rather than increase as a result of the
14 development, we're also going to institute a
15 thorough monitoring plan and adaptive management to
16 ensure that that's the case. We're going to be
17 monitoring the impact on wetlands. The project
18 will adapt the project as needed to ensure that
19 water quality is preserved. It's just not waiting
20 or planning for bad things to happen.

21 The analysis shows that the loads are going to
22 decrease. But we're going to be monitoring so that
23 if something we don't expect does occur, we get an
24 early warning, and we can use our known mitigating
25 actions to ensure that a negative environmental

1 consequence does not occur such as a water quality
2 violation.

3 These monitoring ideas and adaptive management
4 will be discussed now as Christie Kearney comes up
5 for our last portion of the presentation.
6 Christie.

7 MS. CHRISTIE KEARNEY: Thank you.
8 I'd like to take a moment. My name is Christie
9 Kearney. Kearney is K-E-A-R-N-E-Y. I'm going to
10 take a moment to step us back for a moment.

11 So it was recognized in watching our
12 presentation, we have a typo in our presentation.
13 I've talked about -- these numbers on the side are
14 correct overall with the exception of this sulfate
15 total for the 20-year total. That number should be
16 27.6 million kilograms. Still a huge number but
17 it's not billion.

18 So our experts today have explained in great
19 detail the science behind our analysis. It's the
20 details that matter, which the Band has left out of
21 their "will affect" letter or mischaracterized in
22 their presentations yesterday.

23 The understanding of these details is what led
24 the agencies to the issuance of our permits.
25 However, the agencies didn't just rely on the

1 science and our modeling. I'm now going to talk
2 through the monitoring required by our
3 environmental permits, the annual analyses and
4 verification evaluations that we're required to do,
5 and the adaptive management and mitigation laid out
6 in our permits.

7 This slide shows our extensive comprehensive
8 water and wetland monitoring required as a result
9 of our NorthMet permits. This compilation of
10 monitoring required is from our two NPDES permits,
11 our consent decree, our 401 water quality
12 certification, our 404 permit, our Wetland
13 Conservation Act decision, and our permit to mine.

14 This includes stream water quality, stream
15 flow, groundwater quality, and groundwater levels,
16 wetland hydrology, wetland vegetation, wetland
17 water quality, industrial water collection, treated
18 water discharge, and macroinvertebrate and fish
19 monitoring that we're required to do. 280
20 monitoring locations in total.

21 Many of these have been underway throughout
22 the environmental review process, but there's a
23 number of these that are new monitoring locations
24 that have started once our permits were issued.

25 For example, we have 16 years of wetland

1 hydrology data. This creates a robust data set to
2 evaluate potential project impacts from baseline
3 conditions. We're not aware of any other mine that
4 has a monitoring program as robust as this.

5 Now let's focus in on the mercury monitoring
6 since that's what's most important in the
7 presentations yesterday and today.

8 We are and will be monitoring mercury at 66
9 different locations around our project site. In
10 stream water quality, in wetland water quality, in
11 industrial water collection, and in our treated
12 water discharge.

13 The MPCA required monitoring to confirm the
14 expected outcomes of our cross-media work and to
15 ensure the ability to perform adaptive management
16 if changes were found that were attributable to the
17 project. This mercury monitoring is compiled from
18 our two NPDES permits, our 401 water quality
19 certification, and our permit to mine.

20 The Band has contended that there's not enough
21 monitoring to detect harm. This slide and the
22 prior slide showing our 280 monitoring locations
23 shows that that claim is incorrect.

24 In addition to monitoring, the agencies also
25 required -- also included numerous permit

1 conditions that require annual review of our
2 monitoring results. Many of these analyses are
3 listed on this slide. We're required to perform an
4 annual potential indirect wetland impact assessment
5 to evaluate wetland water levels and vegetation.
6 This is from our 404 permit, our 401 water quality
7 certification, and our Wetland Conservation Act
8 decision.

9 We're required to do an annual evaluation of
10 stream and wetland of interest water quality
11 monitoring data to evaluate against our baseline
12 conditions and our cross-media analysis results and
13 predictions based on our 401 water quality
14 certification conditions.

15 We're required to do an annual groundwater
16 evaluation to assess monitoring results, the
17 suitability of our monitoring network, spatial
18 distribution of our groundwater quality, and
19 potential for north flow at the mine site according
20 to our NPDES permit conditions.

21 Our NPDES conditions also require us to do an
22 annual comprehensive performance evaluation to
23 assess the performance of our engineering controls
24 and our monitoring network.

25 And we also have many other annual reviews

1 that we're required to do for our permit to mine
2 and our water appropriation permit that I won't get
3 into today.

4 Additionally, once our monitoring results have
5 been analyzed, we're also required by permit
6 conditions to perform an annual verification,
7 modeling, and evaluation.

8 In this annual assessment we'll be assessing
9 the predictions of our water quality and quantity
10 and comparing them to the actual observed
11 monitoring data. We'll be verifying previously
12 predicted long-term impacts from our EIS and
13 permitting by rerunning our water models with the
14 actual observed data from the monitoring.

15 We're required to determine if changes are
16 needed to remedy unacceptable impacts that might be
17 recognized in the rerunning of our water models or
18 in the monitoring data itself and implement our
19 adaptive management and contingency mitigations
20 that we've already laid out.

21 And every 5 years we're required to
22 reevaluate, rerun our underlying conceptual models
23 such as our MODFLOW model. This is required by our
24 permit to mine, our NPDES permit, and our water
25 appropriation permits.

1 Now let's talk about adaptive management and
2 mitigation because it doesn't appear that that was
3 well understood by the Band based on the discussion
4 yesterday.

5 PolyMet has proposed an adaptive management
6 approach. Adaptive engineering controls can change
7 as a result of monitoring or monitoring data or
8 modeling data. Our water treatment plant is an
9 example of an engineering control. It's designed
10 to be modular so if we're seeing higher flows or
11 higher loads, we can add additional units to it to
12 be able to expand the engineering control and make
13 sure that we're meeting our permit conditions and
14 requirements.

15 Additionally, contingency mitigations have
16 already been laid out in our permitting documents
17 and could be enacted if required.

18 Every one of our major permits includes
19 adaptive management processes and mitigation
20 measures to evaluate and consider.

21 For example, the 404 permit states that when
22 changes are recognized, monitoring report shall
23 include recommendations for appropriate steps to
24 respond to the documented changes to include
25 additional monitoring, adaptive management, and/or

1 compensatory mitigation. Note that it says when
2 changes are recognized, not when permit violations
3 are made. So this is required in addition to our
4 404 permit by our 401 certification, our NPDES
5 permit, our permit to mine, our Wetland
6 Conservation Act decision, and our water
7 appropriation permit.

8 So to wrap up, our project will not affect the
9 quality of the Band's waters so as to 0 any of the
10 Band's water quality requirements.

11 In summary, our project will reduce sulfate
12 and mercury loading and specific conductance in the
13 St. Louis River. This statement was true from the
14 EIS as well as the additional analyses completed
15 for permitting. The Band needs to show that both
16 the EIS and the permitting documents were wrong.
17 Fifteen years worth of analyses to show a violation
18 of their water quality standards. They have not so
19 far. Speculation is not enough to show a violation
20 of a water quality standard.

21 There are adequate controls in place, both in
22 project design and as permit requirements, to
23 ensure that the project will not cause or
24 contribute to a violation of water quality
25 standards for sulfate, mercury, methylmercury, or

1 specific conductance or any other water quality
2 standard at the Fond du Lac Reservation in the
3 lower St. Louis River 116 river miles away.

4 The agencies didn't just rely on the science
5 or our modeling that they reviewed and approved.
6 They put into effect, we have over 7,000 permit
7 conditions that we need to comply with, including
8 comprehensive monitoring, annual verification
9 modeling and evaluation, adaptive management, and
10 contingency and required mitigations.

11 Our project reuses existing infrastructure
12 bringing the site up to modern standards and
13 cleaning up legacy issues in the process, including
14 cleaning up the Embarrass River, the Partridge
15 River, and the St. Louis River as a result of past
16 mining disturbances that have occurred on our site.

17 Currently, we're the only discharge in
18 Minnesota that's required to meet the wild rice
19 standard at the end of our pipe, which will result
20 in a significant reduction in sulfate in the
21 St. Louis River. This project is for the
22 betterment of these streams, including the
23 betterment of the St. Louis River and water quality
24 at the Band's reservation.

25 Regardless, 116 river miles downstream is a

1 long way, as I mentioned earlier. And although the
2 Embarrass River and the Partridge River will
3 clearly show this cleanup, it will be mostly
4 undetected in the St. Louis River at Forbes and at
5 Fond du Lac because our flow is less than 1 percent
6 of the flow at the Fond du Lac Reservation.

7 PolyMet will produce metals that are essential
8 for the U.S. sustainability and energy goals and
9 will be one of the only sources of nickel and
10 cobalt which are essential to battery storage.

11 Our project has gone through extensive joint
12 state and federal environmental review and
13 permitting processes with unprecedented community
14 and tribal involvement.

15 Fond du Lac and the EPA were both cooperating
16 agencies for the supplemental EIS and for the final
17 EIS. And what wasn't mentioned yesterday was that
18 although we did get a failing grade on an earlier
19 version of our EIS, our draft EIS, we went back to
20 the drawing board. We did a supplemental EIS,
21 completely changing our project, including adding
22 the requirement for the 10 mg/L sulfate standard.
23 And as a result, the EPA gave our supplemental
24 draft EIS an EC-2 rating which is the highest
25 rating a mining company has ever gotten in the U.S.

1 And lastly, our project meets the definition
2 of responsible domestic mining called for in the
3 Presidential Decree on the Defense Production Act.

4 I thank you for allowing us the opportunity to
5 present the full story of our project.

6 COLONEL JANSEN: Thank you, Christie
7 and thank you to Steve, Cliff, and Greg for your
8 presentations.

9 Greg, I'll make a note that our YouTube
10 recording cut out at the period of your summary
11 slide. However, we have a backup recording and a
12 backup to the backup recording and a transcript, so
13 we're sure to capture that portion.

14 So now we'll take our recess until 12:30.

15 (A break was had in the proceedings.)

16 COLONEL JANSEN: Welcome back,
17 everybody. We'll move into our afternoon session.
18 Each party has a chance for two-hour rebuttal.
19 We'll have a recess in between.

20 I'll go ahead and recognize Vanessa
21 representing Fond du Lac Band.

22 MS. VANESSA RAY-HODGE: Good
23 afternoon. My name is Vanessa Ray-Hodge, and I am
24 outside counsel for the Fond du Lac Band of Lake
25 Superior Chippewa.

1 In attempting to discount the Band's,
2 well-grounded and scientifically based "will
3 affect" determination and objection, PolyMet
4 continues to try to hide behind smoke and mirrors.
5 Its contention that the Band's reservation is
6 located too far downstream from the project to
7 suffer impacts is simply wrong as laid out in the
8 Band's objection and presented by our experts
9 yesterday.

10 Significantly, EPA agrees with the Band that
11 its downstream reservation waters will be impacted
12 by the proposed project.

13 PolyMet continues to assert that there will be
14 no violations of the Band's water quality
15 standards, but PolyMet ignores the fact that under
16 existing conditions, there are already exceedances
17 of many of the Band's downstream water quality
18 standards, including its numeric standard for
19 mercury and specific conductance.

20 Additional discharges from the proposed
21 project which assume that the state standards will
22 be met have nothing to do with the Band's
23 downstream standards. And as you heard yesterday,
24 PolyMet's conclusions are based on significantly
25 flawed studies which are insufficient to show all

1 hydrologic impact, studies which EPA in its
2 recommendations characterize as extremely cursory.

3 PolyMet also suggests that the Band ignored
4 concepts like alleged reductions in mercury and
5 sulfate due to the proposed project's operations.
6 That is incorrect and our experts will address
7 briefly today those contentions and more fully
8 address those in written comments.

9 Our experts will similarly address PolyMet's
10 absurd conclusion that the Band's objection is
11 speculative and not based on evidence that
12 concretely shows that there will be violations of
13 the Band's downstream water quality standards.

14 A contention by PolyMet that EPA has also
15 implicitly rejected by agreeing with the Band and
16 finding that the Band's objection is,
17 "well-grounded in contemporary scientific
18 research."

19 More broadly, despite PolyMet's allegation
20 that the project has been fully considered and
21 evaluated as part of the EIS process, the Band has
22 been challenging the conclusions in the EIS since
23 the beginning, including the efficacy of things
24 like the seepage capture system, the construction
25 of the tailings basin, water treatment, and more.

1 The Band submitted its comments and concerns on
2 these issues as part of its "will affect"
3 determination.

4 It is true that the Band was a cooperating
5 agency during the environmental review process, but
6 the Band was never respected or listened to as a
7 cooperating agency. In fact, the Band made it very
8 clear during that process and in the federal
9 litigation that the federal agencies wrongly
10 disregarded the Band's expertise and information
11 during that process. And we are here today due to
12 the Band's continuing and ongoing concerns because
13 the federal court in our litigation agreed that EPA
14 failed to issue a "may affect" determination that
15 considered the potential impacts of the proposed
16 project's discharges to the Band's downstream
17 waters and other treaty resources.

18 So this is the first time that both the EPA
19 and the Corps are being required to take a look at
20 the Band's evidence and address the Band's
21 concerns. So to be clear, the EIS is being
22 challenged as part of this process.

23 Moreover, the Corps cannot look to permits
24 issued to PolyMet to provide cover to PolyMet
25 despite their contention that somehow the Band's

1 claims are contradicted by those other permits.

2 It is also important to note that the Band is
3 engaged in ongoing litigation that is far from
4 resolved on many of the State permits issued,
5 including the NPDES or Section 402 permit.

6 In any event, EPA agreed that those permits do
7 not protect or ensure compliance with the Band's
8 downstream reservation water quality requirements.

9 More specifically, EPA agreed that there is a
10 clear violation of the Band's standards because no
11 agency ever made PolyMet comply, for example, with
12 the Band's numeric mercury standard that is twice
13 as low as Minnesota's finding, and I quote, the
14 Clean Water Act Section 402 individual permit
15 authorizes continued exceedance of the Band's water
16 quality standards for mercury because it allows a
17 discharge from the wastewater plant in excess of
18 the Band's water quality standards for mercury of
19 .77 ng/L. And the receiving waters to this
20 discharge within the headwaters of the St. Louis
21 River already exceed the Band's water quality
22 standard for mercury.

23 Under Section 401 of the Clean Water Act, as I
24 mentioned yesterday, the Corps has a statutory
25 obligation to look at and evaluate all discharges

1 from the project and the impacts of those
2 discharges on the Band's downstream waters,
3 including the discharges discussed in the NPDES or
4 402 permit issued by the State and the State's 401
5 certification.

6 So while PolyMet alleges that they are subject
7 to 7,000 permit conditions, importantly, not one of
8 those conditions is keyed to the Band's downstream
9 standards.

10 Additionally, because this matter involves the
11 Fond du Lac Band, a federally-recognized Indian
12 tribe that is a sovereign nation with a
13 government-to-government relationship with the
14 United States, the Corps has an independent legal
15 obligation to look at and evaluate any impacts its
16 action has on the Band's treaty rights pursuant to
17 this process, which PolyMet completely ignores.

18 Similarly, the Corps has an independent
19 responsibility to consider environmental justice
20 issues when it takes agency action.

21 I also want to note for the record that the
22 Band also contends that a dam failure of the
23 tailings basin will lead to violations of the
24 Band's water quality standards.

25 For example, a catastrophic failure will

1 release hundreds of gallons of tailings into the
2 watershed which will flow down the St. Louis River
3 to the Band's reservation. While EPA reviewed the
4 Band's objection on this issue, it ultimately
5 deferred to the Corps on the dam construction
6 analysis.

7 The Band has been challenging the upstream dam
8 construction method for years. And while this
9 matter is also in litigation before the state
10 courts on State mining permits, the Corps has an
11 independent obligation to review the Band's
12 concerns on the tailings basin's dam as part of
13 this process. And, therefore, we formally request
14 that the Corps do that and submit that the Corps
15 cannot rely on the State permits which continue to
16 be tied up in litigation as part of its decision.

17 In sum, the Band's scientifically grounded
18 analysis that show the project's discharges will 0
19 the Band's downstream water quality standards as
20 you heard from the Band yesterday.

21 EPA agrees with the ultimate conclusion of the
22 Band. There are no conditions that could be placed
23 on the suspended 404 permit that would ensure
24 compliance with the Band's downstream water quality
25 standard.

1 PolyMet's presentations which either continue
2 to rely on prior faulty evaluations or add window
3 dressing in attempt to minimize the Band's
4 objection do nothing to change this outcome.

5 Again, PolyMet attempts to characterize their
6 conclusions with certainty but those assertions are
7 based on flawed and inappropriately limited
8 studies.

9 As mentioned by Chairman DuPuis yesterday, the
10 Corps has only two options under Section 401(a)(2).
11 Place conditions on the Section 404 permit that
12 would ensure compliance with the Band's water
13 quality standards or not issue the permit. Only
14 those two options.

15 Based on the evidence before the Corps,
16 including PolyMet's futile attempts today to
17 discount the Band's objection, the Corps cannot
18 reissue the suspended 404 permit and must revoke
19 it.

20 Although we will more fully respond to
21 PolyMet's presentations in writing, our experts
22 will now like to address some of the contentions
23 made by PolyMet today.

24 And I'd first like to call up Nancy Schuldt,
25 the Band's water project coordinator.

1 MS. NANCY SCHULDT: Hello again. I'm
2 Nancy Schuldt, Fond du Lac water projects
3 coordinator.

4 So there were a lot of interesting contentions
5 this morning in the presentation from PolyMet and
6 their experts and consultants. And there's a
7 number of times when I really wanted to jump up and
8 say something but that would not be appropriate.
9 And I appreciate that that didn't happen yesterday
10 when we were speaking about our perspectives.

11 As Vanessa mentioned, despite our engagement
12 in this project throughout the environmental review
13 process and the permit review, as a cooperating
14 agency that involvement and engagement absolutely
15 did not result in a reasonable or fair
16 consideration of what we brought to the table for
17 the co-lead agencies and the permitting agencies to
18 incorporate in the information that the public saw
19 and could respond to and what ultimately formed the
20 basis of the records of decision and the permits
21 that were issued.

22 And there were some comments made earlier this
23 morning about how we didn't, in our presentation
24 yesterday, account for the fact that the project,
25 particularly after the draft EIS was deemed

1 inadequate and unsatisfactory, the supplemental
2 draft environmental impact statement included a lot
3 of changes, a lot of mitigations that were intended
4 to improve the project from an environmental
5 performance standpoint. But from the very
6 beginning we have been really clear about
7 identifying substantive deficiencies in the project
8 and in the studies that underlie the information
9 that was presented in the various EIS chapters and
10 versions.

11 I'm going to touch upon just a couple of
12 examples that I think might help clear some
13 misrepresentations up.

14 The discussion about the seepage capture
15 system at the tailings basin, you know, in our
16 admittedly limited experience, none of us are mine
17 engineers or professionals in the field of mining
18 activities, but we've had to learn an awful lot
19 about what happens at a mine to be able to
20 understand and advocate for the protection of the
21 resources we're concerned about.

22 And what we have seen in action just a few
23 miles away from the project site at another
24 taconite facility is that a seepage capture system
25 that has been in place now for, oh, at least eight

1 or nine years that was touted to be virtually
2 100 percent capture rate and is only performing at
3 about 50 to 60 percent capture rate, that's a real
4 consideration for what is being proposed with the
5 PolyMet project.

6 Like at U.S. Steel Minntac, there's supposed
7 to be a cutoff wall that is keyed into bedrock, and
8 that bedrock is assumed to be a no flow boundary.
9 U.S. Steel, operating in the same kind of landscape
10 and glacially-influenced terrain, was not able to
11 key in their sheet piling to bedrock. They
12 encounter glacial erratics and boulders and there
13 is just certainly no integrity to the wall as it
14 was represented when they were going through
15 permitting.

16 So fundamentally, structurally, that
17 confidence that you can put a cutoff wall from the
18 ground down to bedrock and assume that water is not
19 going to get past it, both through the bedrock, it
20 is not a no flow boundary, there are fractures,
21 that's a misrepresentation of what the reality of
22 this landscape is.

23 But the second element of their seepage
24 capture system which is intended to improve upon
25 that structurally is that they're proposing to

1 install a series, a network of groundwater wells
2 around the perimeter where the cutoff wall is
3 installed. And those wells are intended to pump
4 groundwater sufficiently to create an inward head
5 pressure so that all of the water that is seeping
6 out of the tailings basin already, by design, will
7 be pumped back by those groundwater wells.

8 So I would expect that they will see perhaps a
9 more than 50 percent capture rate, not 100 percent,
10 but more than a 50 percent capture rate because
11 they are literally creating conditions that will
12 keep the water flowing intending to flow backwards
13 towards the tailings basin.

14 But what is not considered or discussed in any
15 way, shape, or form in the environmental review or
16 in any of the documents underlying the permits is
17 that this scenario effectively cuts off the source
18 of waters to the wetlands just outside the tailings
19 basin cutoff wall. And that's a complete complex
20 of wetlands to the north of the tailings basin.
21 And they are completely fully saturated with
22 contaminated tailings water from the former LTV
23 operations. So right now they are fully saturated
24 with highly-contaminated tailings basin water, but
25 they are not going to continue to be replenished as

1 the existing conditions have right now once this
2 cutoff wall and those groundwater pumping wells are
3 operational.

4 So you're going to change that hydrology in
5 those wetlands over time, but you are going to
6 continue to have the release and the northward
7 migration of that plume of contaminates for many
8 years to come. And the simplistic, arithmetic
9 accounting of the improvements that PolyMet asserts
10 that they're going to make in water quality and
11 sulfate loading to the Embarrass River watershed
12 have completely ignored the fact that they're not
13 stopping that massive volume of water and its
14 contaminant load in those existing wetlands as it
15 migrates north to the Embarrass River watershed.

16 So the numbers that they were promoting in
17 terms of the improvements in sulfate loading
18 because of that seepage capture system are in
19 incomplete at best.

20 Over at the mine site in the Partridge River
21 watershed there was much made about our failure to
22 account for how they were going to manage
23 stormwater and mine contact water, that it was
24 going to be captured, collected, transported over
25 to the wastewater treatment plant. In other words,

1 whatever water was contaminated by contact with
2 waste rock or dust or pumped out of the mine, the
3 stormwater that fell within the footprint of the
4 mine site would be collected and treated and
5 essentially cleaned up and that there would be no
6 adverse impacts from the mine site.

7 But there was a stormwater general NPDES
8 permit that was issued for that particular source
9 of water and its pollutant load. And as a general
10 permit, you well know it's not subject to the same
11 kind of public review and comment. We objected to
12 that. We argued that it should have been an
13 individual permit, that there should have been
14 scrutiny and public review but there was not. And
15 because there was not, we didn't get a chance to
16 talk about where we thought the deficiencies were
17 and how that was going to not be sufficient to
18 mitigate for the problems that were going to be
19 caused by stormwater contacting mine waste.

20 There's a lack of liners, for instance, in the
21 detention ponds or underlying the OSLA, the
22 overburden storage and laydown area. And there are
23 sources of mercury and methylmercury from this
24 disturbed landscape and these peatlands that have
25 been excavated for the construction of the mine

1 site itself that are themselves a source of
2 mercury, and any runoff water, contact water from
3 those materials is going to be flushing and
4 releasing mercury and methylmercury.

5 And it isn't just a matter of capturing what
6 is running off on the overland on the surface flow,
7 it's also an issue where it can be in the shallow
8 subsurface flow and escaping the berms that are
9 intended to control the stormwater runoff and
10 exiting into the adjacent wetlands and the adjacent
11 open waters like the Partridge River.

12 So the accounting, again, the simplistic
13 accounting for how their stormwater management was
14 going to control all of the problems over at the
15 mine site has a big gaping hole in it from our
16 perspective.

17 Finally, I would just like to say that
18 generally we never made the argument or the
19 contention that it was the sulfate loading or the
20 mercury loading from the project that were the real
21 issue with regards to compliance with our water
22 quality standard. That was not the primary
23 problem. But that's what PolyMet focused on today
24 in their presentation.

25 Really, it is all about this massive wetland

1 destruction and disturbance to the watershed and
2 the profound hydrologic changes that we've
3 described that will contribute to or exacerbate
4 existing exceedances impairments, exceedances of
5 our water quality standards.

6 And the EIS and the permitting process, no
7 matter how long it took, and I know painfully how
8 long it took, never addressed these known
9 reasonably foreseeable ecosystem processes that
10 increase mercury methylation and subsequently
11 bioaccumulation, and that's at the crux of our
12 concern.

13 So there were some, I would call, incomplete
14 representations of the sources of what we believe
15 would be the exceedances or contribute to the
16 exceedances of the Band's water quality standards.

17 And then I'll let the others follow up with
18 some points about other issues that were raised
19 today. Thank you.

20 MR. ESTEBAN CHIRIBOGA: Hello again.
21 So my name is Esteban Chiriboga. Last name is
22 spelled C-H-I-R-I-B-O-G-A. And I have just a few
23 points to address.

24 As Nancy said, a lot of the point of
25 presentations for Fond du Lac had to do with the

1 scale of land use alterations. And earlier this
2 morning there were some examples and trying to link
3 the proposed PolyMet project with Eagle Mine in
4 Michigan and this is incorrect. This is not an
5 acceptable approach given the difference between
6 these mines.

7 Eagle is a very small mine. It is completely
8 an underground mine. The surface footprint of that
9 mine is a fraction of what PolyMet's would be.
10 Mineral processing is all done off site in a
11 different area located about of 60 miles away.
12 Wetland fill at the Eagle Mine was really less than
13 10 acres. There's really no comparison to what
14 PolyMet would be. And indirect wetland impacts
15 from groundwater drawdown were not a problem or not
16 an issue that that mine had to contend with. It
17 had other issues and other points of discussion.
18 It's not a comparable project to PolyMet.

19 Second point I'd like to make is Mr. Council
20 mentioned that MODFLOW is not a good tool to assess
21 hydrologic impacts to wetlands. I believe the USGS
22 would be surprised to hear that.

23 MODFLOW can and is used to assess impacts to
24 wetlands throughout the country. It has a wetland
25 data package that is used to predict project

1 effects on wetland water levels, things like flow
2 routing, import and export of water in wetlands,
3 evapotranspiration. The list goes on. And Brian,
4 in his presentation, gave the example of the
5 DeBeers Diamond mine that, in fact, uses MODFLOW to
6 predict impacts to surface water features like
7 wetlands.

8 But ultimately, the point was beliefs in
9 misdirection because in my presentation yesterday
10 and throughout the history of our comments, we
11 never suggested that MODFLOW should be the one tool
12 to be used in a quantitative wetland impact
13 assessment. As I said yesterday, the process that
14 federal, state, and tribal agencies that
15 participated in the wetland IAP suggested or
16 brought forward involved use of a groundwater model
17 to determine drawdown, but also using the gathering
18 of additional hydrology information at the site,
19 plant lists, all of the types of information that
20 you would need to link these wetlands and to
21 characterize their connection to groundwater one
22 way or the other and this was not done.

23 The last point I think I'd like to make,
24 Mr. council also referred to GLIFWC's analog
25 analysis. I think all I'll say, and as I said

1 yesterday, is that the USGS groundwater modeling
2 results, the new USGS groundwater model recently
3 completed, supports our contention that the FEIS
4 underpredicts drawdown related at the mine site.

5 And so these -- there were two different
6 methods; our analogue approach and the USGS. They
7 agree on the broad points of impacts, and they were
8 done by two different agencies.

9 That is all I have. Thank you.

10 MR. MATT SCHWEISBERG: Good
11 afternoon. Matt Schweisberg,
12 S-C-H-W-E-I-S-B-E-R-G, with Wetland Strategies and
13 Solutions for the Band.

14 Just a couple of points. One of the first
15 slides that was shown by PolyMet today provided a
16 distance comparison to other features in the
17 watershed. Some weren't in the watershed. They
18 were a bit farther away. Those are all irrelevant.
19 It's like comparing the distance from my heart to
20 my finger or the end of my finger and the
21 capillaries of my finger. That's not the point.
22 The point is that they're inextricably connected as
23 the Fond du Lac Reservation is to where the mine
24 site is via the streams, the wetlands, and the
25 St. Louis River.

1 PolyMet couches several of its arguments by
2 using percentages. Percentages are an
3 inappropriate way to present impacts. It's a
4 common tactic that I've encountered in my 40 plus
5 years of experience numerous times, and it's used
6 to trivialize the appearance of adverse impacts.
7 It's the absolute numbers that really matter here,
8 not the percentages.

9 PolyMet said that removal of wetlands to
10 reduce impacts to -- I'm sorry -- to reduce inputs
11 of methylmercury will be a benefit. It might work
12 that way, but when you think about the bigger
13 picture, that's kind of an absurd argument. It
14 completely ignores all the ecological services and
15 functions of the wetlands that will be removed and
16 that would degrade the designated uses and 0
17 antidegradation of the Band's water quality
18 standard. In particular, fish and wildlife. And
19 it disregards the Band's treaty rights.

20 As Nancy mentioned, the ring of capture
21 system, the ring of wells, capture system wells to
22 prevent migration of contaminated water will
23 actually dewater the wetlands and streams outside
24 that capture system. It's both unrealistic and
25 it's totally unaddressed by PolyMet in its

1 documents.

2 The entire watershed is already Oing water
3 quality standards. That's been mentioned a few
4 times. This is a matter of cumulative impacts.
5 Whether a relatively small percentage or a large
6 percentage, it's still a discharge that will
7 contribute to violations of the Band's water
8 quality standards. That's what really matters.

9 (A break was had in the proceedings.)

10 DR. BRIAN BRANFIREUN: Thank you very
11 much.

12 This morning we saw some very nice graphics.
13 I wish that for my courses that I teach at the
14 undergraduate level that I could produce figures
15 that look like that. I think my students would
16 really appreciate them.

17 I have a number of things that I'll address
18 ultimately in writing on behalf of the Band. There
19 are just a few things that I think are important to
20 highlight today as part of that process.

21 The first is that there was actually a
22 misrepresentation of the way that mercury is
23 delivered to the environment this morning. And
24 it's one that's important because it reveals a lack
25 of contemporary understanding of the processes that

1 are at work here.

2 Mercury in rainfall is, in fact, not the
3 largest source of mercury to watersheds anywhere in
4 the world. Atmospheric gaseous mercury, so the
5 mercury that's in the air around us right now is
6 assimilated by plants directly. Those plants
7 become part of the soil and that actually makes up
8 the largest input of mercury to soils anywhere.
9 And that was in some very important work but Daniel
10 Obrist in 2017 using new techniques to look at
11 natural stable isotopes of mercury in the
12 environment.

13 So what that means is that mercury that is
14 deposited to the landscape in rain isn't directly
15 conveyed to streams and rivers. In fact, the only
16 mercury that's delivered from the atmosphere
17 directly to streams and rivers is the rainfall that
18 falls directly on that water surface, the rain drop
19 directly to the stream or to the lake. The rest of
20 the mercury that falls from the atmosphere is
21 incorporated almost completely into soils, and it's
22 slowly released from that pool to soil water,
23 groundwater, runoff. And that was work what we did
24 in the early 2000s in the Experimental Lakes area
25 which is actually just north of International Falls

1 in Minnesota. It's very similar border landscape.

2 So the mercury that comes from the atmosphere
3 as a gas and is incorporated into plants then gets
4 incorporated into the soils as organic matter. And
5 this large pool of mercury that exists in soils is
6 the main source of mercury to surface waters and
7 streams. And in fact, that pool of mercury is
8 large enough that even if we cut off mercury today
9 from the atmosphere -- and we've done a very good
10 job of that, in fact. I mean, as I discussed
11 yesterday, there's some evidence that in Minnesota
12 and Voyageurs National Park that mercury in rainfall
13 is down 30 some odd percent over the last couple of
14 decades, which is great news. The bad news is that
15 there's probably several hundred years of mercury
16 still in soils to continue to contribute to mercury
17 exceedances in stream waters and lakes in
18 Minnesota.

19 So what that means is that, as Matt just
20 suggested, looking at this as a game of percentages
21 really doesn't matter because there's more than
22 enough mercury in the environment already to
23 methylate, especially in these wetland soils. And
24 the emphasis on the mass balance with precipitation
25 that we saw this morning kind of draws our

1 attention away from the indirect effect of project
2 impacts that we discussed yesterday which is really
3 focusing on hydrologic changes and interactions
4 with soils and this existing pool and the existing
5 condition that Nancy just highlighted.

6 We'll also recall from both Mr. Donohue's and
7 Mr. Twaroski's presentation that there is no plus
8 or minuses on any of those numbers, particularly
9 for mercury.

10 I think it's interesting that when we have the
11 presentation of a change in load does something
12 like the third decimal place in a value in a mass
13 balance. There's a misleading and, in fact,
14 scientifically incorrect implication of precision
15 where real precision is when we actually express
16 uncertainty. What is the plus or minus value
17 associated with that? What does that plus or minus
18 that's both a function of the concentrations and
19 the variability in that concentration in the
20 environment as well as our uncertainty in our
21 hydrologic budget.

22 I would also like to suggest that in this
23 case, and in most cases, there's no consideration
24 of our changing environment, our increasing
25 frequency of wetting and drying extreme events and

1 the potential for flushing events that exceed those
2 which we are currently experiencing.

3 Mr. Donohue had a very nice graphic about
4 where methylation occurs in wetland sediments as
5 well as where the mercury in sulfate comes from.
6 Unfortunately, it was technically a little bit
7 incorrect.

8 He suggested that methylation happens in the
9 wetland sediments, which indeed it does, but that
10 the sulfate and mercury was associated with the
11 mercury and sulfate that's coming from the
12 atmosphere on an annual basis. So that's the
13 diagram here on the left. Of course, I didn't have
14 the luxury of putting together a nice graphic, so
15 it's a little rough.

16 He suggested that the fluctuation of the water
17 table created the conditions that then resulted in
18 the formation of methylmercury as a result of the
19 input of sulfate and mercury from the atmosphere.
20 That's actually technically not correct.

21 As I just suggested, the pool of mercury
22 that's in the soils is plenty. In fact, it's
23 plenty to continue to be methylated for decades or
24 centuries, in fact. And the inputs of sulfate,
25 both from the atmosphere as well as from runoff and

1 additional sources like streams and discharges, are
2 actually the input into the reactor where
3 methylmercury is formed and where sulfate is
4 regenerated. Water table fluctuation influences
5 all of those processes but it's actually a much
6 more complex interaction, and it's one that's in
7 fact driven more strongly by interactions with the
8 catchment hydrology than the input of mercury from
9 the atmosphere, which I think was sort of a bit of
10 a redirect.

11 We also had some discussion about water table
12 fluctuation and how natural fluctuation was the
13 force in action here driving methylation. Indeed
14 it does influence this process. We argued in the
15 "will affect" memo and in previous opinions that
16 I've written that underdrainage amplifies the
17 natural fluctuation that we may expect as a result
18 of both annual variability, as well as climate
19 change-induced increases in fluctuation
20 invariability. It also discounts the fact that the
21 wettest time of year in the spring during snowmelt
22 is when we have our major runoff events in
23 landscapes like northern Minnesota.

24 The frost table in wetlands does, in fact,
25 create a transient perch situation in which the

1 exchange of water and the delivery of mercury
2 sulfate and methylmercury is actually at its peak.
3 We published work to that effect with my own
4 students in the late 1990s. Sorry. The late
5 2009s.

6 There are a number of other issues that I
7 think are important. However, we ended the
8 presentations this morning with some discussion
9 about demethylation. Demethylation is not a
10 process that's going to offset increases in
11 methylation because the concentrations of
12 methylmercury that are in the environment are
13 actually the result of the competitive processes of
14 methylation and demethylation that are happening
15 all the time. So there is no failure to discuss
16 demethylation as a white knight process that might
17 result in the reduction of methylmercury in the
18 environment because it's not something that needs
19 to be discussed separately from the concentrations
20 of mercury that's in the environment.

21 All experimental data that I presented
22 yesterday are concentrations of mercury and
23 methylmercury in the environment. And those
24 concentrations are the net product of the
25 competitive processes of methylation and

1 demethylation. So when we have increases in
2 methylmercury as a result of sulfate, for example,
3 we have the lower case where methylation is
4 enhanced. It's greater than demethylation, so the
5 methylmercury concentrations will increase.

6 We can also have geochemical conditions in
7 which demethylation, the process of demethylation
8 exceeds methylation and methylmercury
9 concentrations will go down. But everything that
10 we presented yesterday is consistent with our
11 scientific consensus that when we add sulfate, we
12 increase the absolute concentrations of
13 methylmercury in the environment and that includes
14 processes of demethylation which are also ongoing
15 at the same time. It just means that methylmercury
16 is winning under those circumstances. And
17 methylmercury or methylation, the process of
18 methylation wins when extra sulfate is added
19 because it fuels the sulfate-reducing bacteria that
20 are the primary methylators of mercury in the
21 environment.

22 Sulfate-reducing bacteria are not one
23 organism. They are a group, a complex group of
24 bacteria. And indeed, as stated this morning,
25 there are some sulfate-reducing bacteria that are

1 also capable of demethylating. They are not the
2 same organisms that methylate. They're a different
3 group of sulfate reducers.

4 We also know that there are many other
5 organisms that also methylate mercury. However, in
6 systems like fresh water systems like northern
7 Minnesota, the sulfate-reducing bacteria -- species
8 of sulfate-reducing bacteria are the dominant
9 methylators.

10 So what we measure in the environment we can
11 measure methylation and demethylation together.
12 It's much more complicated. We have to use stable
13 isotopes of mercury to do it. But really, what
14 we're interested in is the net outcome. And the
15 net outcome of adding sulfate is increased
16 methylation which results in a net increase in
17 methylmercury concentrations in the environment.

18 I'd like to end with this final reference to
19 this manuscript, which I would expect everyone is
20 fairly common -- fairly confident that I would
21 comment on since I'm a coauthor of this paper from
22 2012. This comment was used to support the
23 contention that demethylation will somehow be a
24 mechanism that will reduce mercury in the
25 environment. Some sort of remediation approach.

1 This statement was both selected quite specifically
2 and also redacted in a way which quite
3 significantly changed its meaning.

4 The quote on the left is the one that was from
5 the presentation and submission to the Corps. That
6 says that "The finding that most of the
7 methylmercury lost, redacted, was likely due to
8 in situ demethylation rather than export from the
9 system implies that the majority of methylmercury
10 produced in response to elevated sulfate deposition
11 may not be transported to downstream aquatic
12 systems."

13 This implies that methylmercury that's
14 produced in wetlands may never get out, may never
15 leave. It would be wonderful if that was the case
16 because we wouldn't have a surface water
17 methylmercury problem if that was true.

18 The full quote from the manuscript which, not
19 surprisingly, I had on my laptop. It was easy for
20 me to find because I was the coauthor of it, is
21 that: The finding that most of the methylmercury
22 lost from the recovery treatment -- because this
23 was a sulfur reduction experiment -- was likely due
24 to in situ demethylation...

25 The following sentence is perhaps the most

1 important. "This is supported by the finding that
2 peat and porewater methylmercury increased by four
3 times in response to the four times increase in
4 sulfate deposition."

5 So the paper demonstrated a proportional
6 increase in wetland methylmercury to sulfate
7 inputs. The qualifying statement is: "but
8 methylmercury flux from the wetland" -- so the
9 methylmercury lost from the wetland -- "in the
10 first year of this study was only two times." So
11 we had a 400 percent increase in sulfate loading, a
12 400 percent increase in methylmercury in
13 porewaters, but only a 200 percent increase in
14 export. 200 percent increase is still a
15 significant increase. The mass balance, if we
16 actually would like to use that term appropriately
17 here, is that we had a 200 percent increase in
18 porewaters that we couldn't account for.
19 Naturally, demethylation is a process that's
20 happening.

21 This quote is misused in this case to imply
22 that demethylation is going to remove mercury and
23 methylmercury from the system and prevent export to
24 downstream waters which, in fact, is most certainly
25 not the case and certainly misrepresents the

1 finding of this paper, which I felt important to
2 present clearly here for the record.

3 Thank you.

4 I'll restate that. I did forget to make one
5 point that I will.

6 So I think our representation has given you
7 sort of a slate of people with the longest and most
8 complicated names for the record that I apologize
9 for. Brian Branfireun.

10 The final point that I would just like to make
11 is that the word speculation came up a lot in a lot
12 of the written submissions as well as the
13 presentations over the last day.

14 To me, as an academic, speculation has very
15 specific connotations. Speculation is usually --
16 it certainly implies, if not directly, means an
17 unfounded conjecture, some sort of a statement that
18 isn't based on fact or preexisting knowledge. I
19 would prefer to use the term conceptual model or
20 hypothesis for the kinds of work that we have
21 presented and that was supported by the EPA.
22 Speculation implies that there isn't a scientific
23 basis, which is not true.

24 A conceptual model or hypothesis is based on
25 the knowledge that we have in our contemporary

1 scientific understanding of the way the world
2 works. It also implies something that's testable.
3 It implies a collection of measurable parameters
4 that could be evaluated for their relative
5 importance.

6 I believe that what's been put forward from
7 the Band is a conceptual model, a series of
8 hypothesis that are based on sound scientific
9 evidence and fact based on other work that's been
10 done in our discipline and other disciplines
11 related. And a sound conceptual model that
12 incorporates both direct and indirect effects
13 associated with the proposed development is
14 something that hasn't been done.

15 The only discussion that we heard today was
16 about direct effects. We didn't hear about
17 indirect effects of wetlands. We saw minimization
18 about the indirect effects in the area of drawdown
19 and none of the mercury or sulfate processes
20 associated with those have been considered. And
21 despite the discussion about methylmercury this
22 morning, there has been effectively no discussion
23 about the processes of methylmercury and
24 methylation in the environment in the proximal
25 regions associated with the development.

1 Thank you.

2 MR. THOMAS HOWES: Good afternoon.
3 Thomas Howes, Fond du Lac Band's natural resource
4 manager. Last name H-O-W-E-S, since we're using
5 English today for the most part.

6 I guess I just want to sum up and end, you
7 know, our address to the Corps and for, you know,
8 purposes of this rebuttal part of the hearing and
9 kind of going back to things that I discussed
10 yesterday and things that others have brought up is
11 about our relationships.

12 Federal agencies and the federal government
13 have a broader responsibility to tribes, and in
14 particular in this case to the Fond du Lac Band and
15 by extension to myself and the 4800 other Fond du
16 Lac Band members. And that extends to the way that
17 you -- your agency takes action as it pertains to
18 our treaties and not just individual permits.

19 Treaties are the supreme law of the land, in
20 my understanding. I think you can find those that
21 corroborate that. And so this isn't exclusively
22 about compliance with the Band's water quality
23 standards but also the broader responsibility or
24 the sort of dual responsibility that federal
25 agencies have to the Band and to all tribes.

1 So in my mind permitting decisions that
2 abrogate the treaty, that degrade our rights that
3 we've retained in those treaties shouldn't be
4 allowed. It's a violation of the federal
5 government's trust responsibility.

6 Beyond that, our sitting president has issued
7 an executive order on environmental justice. We
8 feel that allowing this permit to stand is a
9 violation of our standing as people, that it's an
10 issue of environmental justice.

11 These wetlands that we're discussing in great
12 detail by people far more qualified than myself to
13 discuss are extremely valuable. And that's
14 probably the thing that I want to leave you with is
15 these areas will be gone forever. And, you know,
16 we talked about wetland ecology and the values that
17 those things have but there's also the wildlife
18 connectivity and the plants and the diversity that
19 those places support. Those are treaty rights that
20 we retained and that we expect federal government
21 to uphold.

22 So I guess I would just remind you yesterday
23 of the Chairman's request since he can't be here
24 and just reiterate that. I think Vanessa made that
25 clear. I appreciate your time and your attention.

1 It's a lot of information in a short period of
2 time. So migwetch for your understanding. Thank
3 you for your understanding.

4 MS. VANESSA RAY-HODGE: Thank you to
5 all the Band's experts. We appreciate all the hard
6 work that you guys have done throughout this
7 process.

8 Thank you, Colonel Jansen for your
9 participation and hearing us out the last couple
10 days. That concludes the Band's rebuttal
11 presentation, so I'll turn it over to you.

12 COLONEL JANSEN: Thank you all for
13 your final statements. I appreciate that. We're a
14 little bit ahead of our normal schedule. I think
15 we're able to compress our afternoon schedule, so
16 we'll move forward with a 30-minute recess. We'll
17 return at 2:05 and then we'll proceed with
18 PolyMet's rebuttal. Thank you.

19 (A break was had in the proceedings.)

20 COLONEL JANSEN: Welcome back
21 everybody. We'll resume our hearing and I'll
22 recognize Ms. Christie Kearney for PolyMet 2-hour
23 rebuttal, which is our final event of the day.

24 MS. CHRISTIE KEARNEY: Thank you for
25 this opportunity to say a few more words about what

1 was just said.

2 So I want to start with the fact that the
3 Band's claims have been extensively studied and
4 resolved. The Band has been a cooperating agency
5 throughout the environmental review process, and
6 they provided a lot of feedback throughout the
7 environmental review and permitting. Many of our
8 expanded studies are specifically to address the
9 comments that they raised.

10 As mentioned earlier, our project completely
11 changed between the draft EIS and the supplemental
12 draft EIS. We took a step back, reassessed our
13 project from the get go.

14 Some major changes that happened since that
15 time as a result of the process that we had just
16 gone through on the draft EIS was that we added
17 that containment wall around the tailings basin to
18 capture all of the water from the tailings basin.

19 Early on in the draft EIS, our earlier
20 project, we had a series of interception wells that
21 just pumped the water out of that area. We
22 realized that really to effectively capture all
23 that flow, we needed a more robust system, so we
24 designed the seepage containment system to do that.
25 And I'll get into more detail on that as we go.

1 We also added our membrane treatment plant in
2 order to meet the sulfate standard of 10 mg/L. We
3 needed a system that could really remove a lot of
4 constituents from the water. There's quite a bit
5 of constituents that's coming out of the tailings
6 basin now, and in order to meet water quality
7 standards at the project boundaries, we needed to
8 reduce those parameters and constituents of concern
9 significantly, which we added into our project.

10 And then lastly, the other major change that
11 we made to our project was that we took all of the
12 waste rock that could have potential acid rock
13 drainage and backfilled that into our pit at the
14 end of our project so that wouldn't be a concern
15 long term.

16 As a result of our significant changes to our
17 project, our supplemental draft EIS got an EC2
18 rating from the EPA. This is significant because,
19 as was mentioned, our earlier project got an EU2
20 rating, environmentally unsatisfactory.

21 EC2 rating, environmental concerns of 2 is the
22 highest rating that a mine mining project has ever
23 received in the U.S. It's the same rating as the
24 St. Croix Bridge project and the St. Paul to
25 Minneapolis light rail project.

1 I just want to go through a few of the
2 comments that were raised by the EPA throughout our
3 process. So this is the same letter that that
4 rating was part of.

5 We appreciate the extensive improvements to
6 the project and the clarity and completeness of the
7 environmental review that are reflected in the
8 supplemental draft EIS.

9 The FEIS refined the quoted statement to more
10 clearly characterize the risks associated with
11 mercury releases. Based on this risk
12 characterization, the FEIS should explain what has
13 been and will be done to avoid, minimize, and
14 mitigate the mercury releases from the project.
15 That was March of 2014.

16 In August of 2015 another letter came out from
17 the EPA. The PFEIS which is the preliminary final
18 EIS that was circulated for review by the agencies,
19 the cooperating agencies, and the consultants on
20 the project to review it before it's finalized,
21 they stated that the PFEIS reflects many
22 improvements to the project and to the clarity and
23 completeness of the environmental review. Our
24 extensive discussions with the co-lead and
25 cooperating agencies have helped to resolve

1 virtually all of our previous comments.

2 Next the co-leads received a letter in
3 December of 2015 when the FEIS came out. The FEIS
4 adequately resolves EPA's comments on the
5 preliminary FEIS pertaining to base flow and
6 cumulative impacts, model calibration, and
7 contradictory information. Ultimately, the FEIS
8 found no exceedances of the Band's mercury
9 standard.

10 The FEIS statement was that the net effect of
11 these project changes would be an overall reduction
12 in mercury loadings to the downstream St. Louis
13 River upstream of the Fond du Lac Reservation
14 boundary. Therefore, the NorthMet project's
15 proposed action would not add any potential
16 exceedance of the Fond du Lac mercury water quality
17 standard of 0.77 ng/L within the reservation.

18 So the FEIS stated that there would be no
19 exceedance of the Band's mercury water quality
20 standard within their reservation.

21 The Corps of Engineers' record of decision
22 came out and also found that there would be no
23 exceedance of the Band's mercury standard.

24 The net effect would be an overall reduction
25 in mercury loadings to the downstream St. Louis

1 River upstream of the Fond du Lac Reservation
2 boundary. The project is not expected to add any
3 potential exceedance of the Fond du Lac mercury
4 water quality standard of 0.77 ng/L within the
5 reservation.

6 Additionally, there would be no expected
7 change in mercury fish concentrations and no
8 substantial change in human health risks related to
9 fish consumption.

10 Notably, that record of decision came out
11 after the MPCA's 401 water quality certification.
12 This is the fact sheet that came out with this,
13 with the findings. This 401 certification fact
14 sheet is really based on the more extensive
15 cross-media analysis that we've talked about a
16 little bit today that took into account not only
17 our water discharges and our water management
18 actions, but also took into account our air
19 emissions that could potentially affect water
20 quality as well, which is why it's called the
21 cross-media analysis.

22 That fact sheet states that based on the
23 cross-media analysis, the PCA concludes that the
24 project would not result in any measurable change
25 to water quality downstream of the project in the

1 St. Louis River, including downstream locations at
2 Forbes and the upper St. Louis River.

3 It's been mentioned that our NPDES permit was
4 just through the court system. The Court of
5 Appeals did uphold our MPDS permit that it would
6 comply with the Band's water quality standards.
7 The Minnesota Court of Appeals concluded that
8 PolyMet's MPDS permit, 402 permit would comply with
9 the Band's water quality standards because
10 discharges from the project will not alter the
11 waters within the Band's Reservation boundaries.
12 And the permit ensures compliance with the Band's
13 water quality standards.

14 I want to talk a little bit about the seepage
15 containment system because there does seem to be
16 some confusion still on that system.

17 So our cutoff wall surrounds the tailings
18 basin, and it is keyed into bedrock. This cutoff
19 wall was put into our project -- designed as part
20 of our project after the supplemental draft for the
21 supplemental EIS when we took a step back looking
22 at our project. There's several different features
23 of this cutoff wall that I want to point out.

24 We do have a continuous pipeline that follows
25 the cutoff wall around the perimeter of the

1 tailings basin with lift stations. I believe
2 there's three lift stations with pumping systems
3 that will pump that water out. The result of that
4 pumping system will be that it will maintain an
5 inward gradient to make sure that we're capturing
6 all of the seepage in that area.

7 We have extensively modeled the system. It is
8 keyed into bedrock, but we have modeled it through
9 the bedrock. Although this figure says it assumes
10 a no-flow boundary, we did model it through
11 bedrock. There is very little seepage that could
12 escape from this. The EIS assumed 93 percent
13 capture. There are a lot of different designs of
14 cutoff walls whether it's sheet pile wall, a soil
15 bentonite wall, as this is intended to be, and it's
16 all the design that matters with the these systems.

17 The Minntac cutoff wall was brought up today
18 as an example of what this might be. The Minntac
19 cutoff wall was designed to be essentially a French
20 drain system to capture the water from the surface
21 aquifer. It wasn't intended to capture 100 percent
22 of the flow.

23 Another feature of this containment system,
24 while we had 93 percent capture assumed in the
25 environmental review process, we have permit

1 conditions that require it to be better than that.

2 Before I get into those permit conditions, I
3 want to point out we do have -- when we proposed
4 this system, we submitted a memo to the co-lead
5 agencies on the degree of use of these type of
6 systems in the industry. I have several listed
7 here. This memo included a list of about 15 and
8 had a number of design guidance documents that were
9 referenced for the design of these systems. One
10 was the Corps of Engineer's design guidance, one
11 was the Bureau of Land Reclamation guidance. As I
12 said, these systems have been used around the world
13 for decades.

14 I did mention earlier that we have over 7,000
15 total permit conditions. I have a few listed on
16 here from our NPDES permits specifically to the
17 containment system design that PolyMet must
18 construct a permeability cutoff wall keyed into
19 bedrock with collection and capability of removing
20 collected water to the treatment system or the
21 tailings basin. We have to maintain a system of
22 paired monitoring wells and paired piezometers.

23 The reason we have those paired systems is so
24 that we can make sure that we have -- we are
25 maintaining that inward gradient between the

1 outside and inside of the cutoff wall. If we're
2 seeing higher water table on the outside of the
3 cutoff wall, we know we have inward gradient.

4 PolyMet must maintain an inward hydraulic
5 gradient across that system. And if necessary,
6 PolyMet must immediately commence mitigation
7 measures, including sampling, inspection,
8 assessment, pumping, removal, repairs, and
9 upgrades. There's a whole slew of different things
10 in our permit conditions specific to design and
11 operation of this system to verify that it's
12 operating appropriately.

13 I also want to touch briefly on our membrane
14 treatment system, our best available technology for
15 water treatment. I did mention in my earlier slide
16 that the Eagle Mine in Michigan uses this
17 technology. They use reverse osmosis as their
18 primary means of removal chosen as the best
19 available technology, and it was mercury that they
20 were concerned specifically for treatment. They
21 have an RO system at both their Eagle Mine and at
22 their mill for water treatment.

23 When these systems were put in, when that
24 project was proposed, the opponents of the Eagle
25 Mine claimed that it wouldn't work and that it was

1 unproven technology. We do have a lot of data that
2 they have shared with us. We actually used this
3 data in our permitting process because Eagle Mine's
4 been operating for several years with actual data
5 to show the successful removal of mercury.

6 And this is a slide that I used during our
7 permitting process in a meeting with the EPA and
8 the PCA when the EPA expressed concerns about
9 mercury removal from our RO system. This is
10 showing about three and a half years of data. They
11 do have more data since then, and I haven't reached
12 out to them yet to get it but I plan to for our
13 written comment period. But this is the data that
14 we used in permitting and I thought was appropriate
15 to share here.

16 The red line is the influent water coming into
17 the water treatment plant and the blue line is the
18 effluent, so what was treated and discharged.

19 You can see there's one data point that was
20 above the detection limit in their three and a half
21 years of treatment and sampling and discharge. All
22 of those other points are at the 0.5 ng/L detection
23 level. So it is a proven technology and have data
24 to support that.

25 Now I'll turn it over to Steve.

1 MR. STEVE DONOHUE: Good afternoon
2 again. My name is Steve Donohue. Last name is
3 spelled D-O-N-O-H-U-E. Spoke this morning so I
4 won't get involved into the details on my
5 background or anything.

6 I want to provide just a few rebuttal and
7 concluding comments as it relates to some of the
8 water management features of this project. There's
9 been a lot of discussion, a lot of detail about
10 water management, water treatment containment, all
11 this kind of stuff.

12 I want to try to boil that down for the
13 hearing here into some very simple concepts that
14 are really at play here to try to help provide some
15 clarity to what we're talking about here.

16 So I'm going to begin with this graphic. It's
17 just a conceptual schematic of what the project
18 looks like today. We know we have there is we have
19 a tailings basin. We know this tailings basin is
20 seeping water out into these riparian wetlands
21 around the streams that drain into the Embarrass
22 River.

23 The seepage water from the basin is high in
24 sulfate, 300, 400 mg/L, and it's very high in
25 specific conductance. This tailings basin at the

1 plant site are subject to a consent decree to
2 actually clean this facility up. What this project
3 is really about is a brownfield redevelopment
4 effort.

5 We've done this before at the Eagle Mine that
6 Christie just mentioned. The mill processing site
7 which is about 55 miles to the south of the mine is
8 a refurbished brownfield redevelopment project.
9 They have a mill there. They have a tailings basin
10 that was in an iron pit lake that was seeping
11 contaminated water out into a stream that drained
12 into our river like the St. Louis River that was
13 all cleaned up as part of the re-permitting effort.
14 That's really what's being proposed here.

15 The other component of this project is the
16 undeveloped mine site. What's going on there today
17 is we have precipitation going onto the site. Some
18 of that precip goes in the groundwater. It drains
19 into the Partridge River. Some of that water runs
20 off and drains into the Partridge River.

21 At the plant site we also have stormwater
22 runoff from the plant site that drains down either
23 into the Embarrass River or makes its way into the
24 Partridge River. So this is a brownfield that is
25 currently impacting water quality in the upper

1 parts of the St. Louis River watershed.

2 We're now going to look at what's actually
3 taking place, in a nutshell, once this project goes
4 into operation.

5 There's a number of very basic features here
6 that are going to improve water quality just like
7 you would do if you were remediating a site. Many
8 of us who have been in the industry for a long time
9 have been involved in remediation projects where we
10 have waste sites that are leaching chemicals into
11 the groundwater, into the surface water. We've
12 applied all sorts of different types of engineering
13 techniques to clean those sites up. And that's
14 really what's being deployed here at the tailings
15 basin.

16 The first thing that we have to do to clean
17 this site up per the consent decree is to contain
18 this water that's seeping out of the tailings
19 basin. The way to do that is put in this
20 containment wall with, in essence, a French drain
21 on the inside of the containment wall to pump water
22 out, and route that through a water treatment
23 system and then return that clean water to the
24 environment just like you would do for any kind of
25 a remediation project. That's all that's really

1 happening at this tailings basin during operation.

2 These containment features, the slurry wall,
3 very conventional technology. Christie mentioned
4 the Eagle Mine a little earlier. The tailings
5 basin at the mill site actually contains a cutoff
6 wall, slurry wall that goes down about 75 feet.
7 It's keyed into bedrock and that's to hold the
8 contaminated pit water back from seeping out into a
9 wetland and a stream that drains into the Escanaba
10 River.

11 So it's the same type of technology that's
12 being used at that project is being employed in the
13 perimeter of this tailings basin. You can go up
14 there today and do a tour of the site and see how
15 it's working. It's been a very successful
16 brownfield redevelopment. It's had tremendous
17 benefits to the community in terms of employment in
18 allowing that one project to go forward. And as
19 Christie mentioned yesterday, it's the only nickel
20 producing mine in the U.S. right now. Tremendous
21 site.

22 Other features of this project are this water
23 that is collected out of this treatment -- or out
24 of this containment system is going to be routed
25 through a wastewater treatment plant. As Christie

1 mentioned, this is going to employ membrane
2 technology like reverse osmosis. It's been used in
3 many operations. It's used in many industries,
4 pharmaceutical industries, any industry where you
5 have to generate really high quality water or you
6 have to discharge water into an environment where
7 you have very stringent environmental standards
8 like we do in this particular project.

9 So this technology has been employed. The
10 Eagle project, as Christie mentioned, there was a
11 very lengthy contested case hearing. The opponents
12 of the project brought in witnesses that said it
13 was unproven technology. They testified to that.
14 We were involved in that contested case hearing and
15 personally testified in it. There was a professor
16 that was brought in from the University of Nevada
17 Reno who claimed that technology wouldn't work.
18 It's now up and operating. And as the data that
19 Christie showed, it's working as designed. And it
20 is protecting the environment, which we all want to
21 do here. So that's proven technology.

22 The other aspects of this project relate to
23 the mine site. Here at the mine site we have two
24 types of water that are going to be generated. I
25 like to characterize it as contact water so that is

1 water that is coming in contact with the mine
2 pumped out of the pit. It's water that is
3 generated from runoff from the haul roads where it
4 could pick up constituents from the mining
5 operation. So you don't want to let that run off
6 into the environment. All that contact water that
7 is picking up regulated pollutants is routed to
8 mine retention basins and then pumped over to the
9 wastewater treatment plant. It's either reused in
10 the -- that water is either reused in the mill
11 operation or it goes through the water treatment
12 plant.

13 All that water that is treated is then used to
14 augment the wetlands and in the streams around the
15 perimeter of the tailings facility. So we're
16 balancing the system back out. Yes, we're pulling
17 water out of the tailings facility, but we're
18 treating it, and by reintroducing that clean water
19 into these streams through these wetlands, we're
20 balancing the hydrologic system back out with clean
21 purified water.

22 The only other sources of water to the
23 environment are stormwater that's running off of
24 the mine site. This is what I generally refer to
25 as noncontact water, so it's like natural

1 stormwater. It's not coming in contact with any of
2 the mine materials. It's not picking up any
3 regulated pollutant.

4 Same thing at the plant site. There will be
5 stormwater runoff from there. All of that
6 stormwater runoff is routed through sedimentation
7 basin so you're not loading that runoff with
8 sediment that's made its way into the system. Very
9 conventional stuff that's employed at many
10 industries.

11 So at the end of the day what we're really
12 talking about here and why we get the reduction in
13 contaminate loading into the system and reduction
14 in sulfate, the reduction in mercury, the reduction
15 in specific conductivity is because we are cleaning
16 up this site and reducing contaminate loading
17 that's going into the system.

18 If you weren't doing a mine, if we weren't
19 employing and developing an open pit mine here and
20 you wanted to clean this site up, you'd be doing
21 the same thing that is being proposed by PolyMet.

22 The logic that we've heard over the last
23 couple days almost indicates that it's impossible
24 to clean the site up because by cleaning it, we're
25 going to increase loading to the system, and that

1 really defies logic.

2 So I think another way of looking at this
3 project is because it is a brownfield site, there's
4 actually environmental benefits that will be
5 derived from this mining operation.

6 First of all, I think it is fair to say that
7 through the presentations we've seen over the last
8 day and a half that the Band really ignores key
9 water management features of this project. There
10 will be a reduction in mercury due to these water
11 management features and the water treatment system.
12 That's real reduction in mercury that's going into
13 the environment.

14 There will be a reduction in sulfate loading
15 to the environment. There will be a reduction in
16 sulfate loading to those wetlands north and east of
17 the tailings facility that will have a benefit on
18 reducing the amount of methylmercury that's formed
19 in those wetlands. And that's to the tune of
20 1,380,000 kilograms of sulfate that's removed from
21 this area on an annual basis. That's a project
22 improvement. It's an environmental benefit.

23 Increases in mercury sulfate and specific
24 conductance, as alleged in the Band's "will affect"
25 letter will not happen because of the engineering

1 features that are built into this project.

2 Finally, I'd like to just draw a little bit of
3 a comparison for everybody in terms of a nanogram.
4 We talk about 5.2 grams of mercury being pulled out
5 of the system on an annual basis. And on one level
6 a gram, 5.2 grams, doesn't sound like a lot. But I
7 think another way to look at it is it is a pretty
8 significant reduction because the standards that
9 we're talking about as it relates to mercury in the
10 water is that nanogram per liter. So a nanogram is
11 one billionth of a gram. Another way to visualize
12 that is it's the equivalent of a one-pound coffee
13 bag. If A nanogram was one pound of coffee,
14 what's -- what's the -- what would be the weight of
15 a gram? Well, the comparison, an analogy is it's
16 the combined weight of more than 2,470 Boeing 747
17 planes. So there's quite a bit of difference there
18 between an nanogram and a gram when we pull out 5.2
19 grams out of the system, that's pretty significant
20 reduction to the environment for this project.

21 So with that, I'll conclude my remarks and
22 hand it over to Greg Council.

23 MR. GREG COUNCIL: Thank you. Good
24 afternoon again, everyone. My name is Greg
25 Council. I'm back up to talk again about what is

1 going on and give a few clarifications.

2 We heard yesterday about an analysis that was
3 new and I hadn't -- I don't think it had been
4 previously explained in the "will affect" letter,
5 but it was a modeling analysis that showed drawdown
6 contours and it was referred to as USGS GLIFWC
7 analysis. And this is a report that was noted in
8 the slides yesterday where to get that and then
9 this is the actual record. I hadn't looked at it
10 before about. It does say this is a USGS report
11 and that it was prepared in cooperation with
12 GLIFWC. So it's a GLIFWC USGS report. I believe
13 this is the one that Mr. Chiriboga referred to
14 yesterday and today.

15 I just want to point out if you read this, it
16 tells you exactly what the purpose of the model is.
17 And it tells you also what the purpose of the model
18 is not. Specifically, importantly, it says that
19 the model scenarios in the report, the model mining
20 scenarios were not designed to predict effects from
21 any specific future mine, including PolyMet, within
22 this basin. They predicted several mines. None of
23 them were supposed to be used. This is not a
24 prediction model. That is not what the purpose of
25 the model was for.

1 It goes on to say what you have to do to make
2 it for that purpose, but clearly the study itself
3 done by USGS was not intended to be used to predict
4 the effect of the PolyMet mine.

5 Furthermore, if you dig into the details,
6 you'll see that the wetlands were not simulated
7 with any kind of wetland package. They were
8 simulated just standard MODFLOW tools as a standard
9 boundary condition, which actually does limit the
10 amount of infiltration that wetlands are allowed to
11 provide. And what that does, actually -- go
12 through the long explanation. It actually shows --
13 I can show you that that actually limits the amount
14 of drawdown that the MODFLOW model will produce at
15 wetlands. So the groundwater drawdown that's
16 produced is limited.

17 Furthermore, what it does show you is
18 groundwater drawdown in the mine, even though it
19 shouldn't be used for that purpose, it was shown
20 for that purpose, and I just want to point out that
21 groundwater drawdown is not the same as the water
22 level desaturation level in a wetland. They're
23 related but they're not the same. And it's not at
24 all easy to translate, especially because you
25 ignore -- if you ignore, as this analysis did, the

1 entire less permeable conductance of the wetland
2 sediments.

3 I'll stick by the standard -- by the
4 statement, rather, that MODFLOW by itself is not a
5 great tool. It's not an appropriate tool for
6 assessing wetland impacts. Yes, you can use
7 MODFLOW with some other analyses to do so but
8 MODFLOW by itself is not an appropriate tool to
9 assess wetland impacts.

10 One more point is that the USGS model is a
11 steady state model. So rather than predicting what
12 happens on a transient yearly basis like the graph
13 I showed earlier today that showed the mining flow,
14 for instance, as a function of a mining year. The
15 MODFLOW model in the USGS report, USGS GLIFWC model
16 is steady state. It's just a worst case. We're
17 going to assume the mine goes in all at once and
18 see what the maximum impact would be.

19 Yesterday we also heard a little bit about the
20 method that GLIFWC suggested should be used rather
21 than what was used for assessing impacts to
22 wetlands using MODFLOW. We had some other -- I
23 think Mr. Chiriboga pointed out some other
24 assessment called the Crandon Method. This refers
25 to, I believe, a method that was applied in the

1 '90s. I was actually one of the modelers involved
2 in the Crandon project back in the '90s. I
3 remember most of this. It was a while ago. But at
4 that time we used MODFLOW to help understand
5 drawdown around a mine that had been proposed.
6 It's a little better tool in that situation because
7 in Crandon, unlike here, you had thicker
8 unconsolidated, so the drawdown there is little
9 more closely related to what you can expect a
10 wetland drawdown to be.

11 But it wasn't just that. You had to also
12 assess the wetlands to some degree and figure out
13 which ones are precipitation dominated and which
14 ones are groundwater dominated. The groundwater
15 dominated ones are more likely to be impacted by
16 drawdown.

17 And then look at the water budget. Remember I
18 looked at the water budget when we talked this
19 morning. Look at the water budget. Assess the
20 vegetation type and try to figure out what the
21 impacts will be in that way.

22 Now, for the Crandon project there was all
23 separate models according to which model was good
24 enough. Do we need to create another model? And
25 it actually just added a lot of delay because they

1 were competing models at the end of the day.

2 Ultimately, there was never a formal agreement on
3 what would constitute the proper method to define
4 what a wetland impact would be. So the Crandon
5 Method didn't work, at least not for the Crandon
6 project, to the best of my knowledge.

7 Furthermore, the technical part of this, first
8 figure out where did we expect drawdown would
9 likely occur. Then using the information about the
10 wetlands, figure out which of those wetlands are
11 more likely to be impacted by the drawdown. That
12 technical piece is the process that was followed by
13 PolyMet. The only difference is that the drawdown
14 assessment was done using data in an analog site as
15 opposed to a MODFLOW model.

16 So lastly, I'll say that the drawdown impacts,
17 just to reiterate from this morning, they're likely
18 to be somewhat -- we agree that drawdown decreases
19 as you move away from the mine. The Band, of
20 course, suggests that the drawdown should be much
21 larger.

22 This issue was specifically addressed in the
23 FEIS. One of the results, one of the changes or
24 one of the -- yeah. One of the changes in the FEIS
25 relative to prior versions of the document is that

1 they -- in a certain drawdown zone they change the
2 definition of unimpacted wetlands to low likelihood
3 of impact on wetlands based on a reassessment based
4 on some of the Band's comments of how certain types
5 of wetlands behaved.

6 Still, they use distance zones based on
7 observed effects. And I like that because it shows
8 a preference for data over a model.

9 Importantly, the PolyMet mine, as I mentioned
10 this morning, will be in a much less permeable rock
11 formation than steel mine which was the analog
12 model used to pull the data that used to assess how
13 much drawdown occurred.

14 Finally, predicted mine inflows. Again, 1
15 CFS. These are a small percentage of the
16 sub-watershed basin for the Partridge River which
17 means that the effects are not expected to be huge.
18 MODFLOW, while it's a good tool for predicting
19 these kinds of things like mine inflow, it's not a
20 good tool for predicting the level of desaturation
21 from the wetland.

22 One more slide I want to draw on something
23 that's monitor and the effects of drawdown can be
24 mitigated. I want to point out that monitoring
25 water levels is a fairly simple, straightforward

1 exercise and it's being done. It's being done now
2 and will continue to be to be after mine
3 development.

4 If you see drawdown that you weren't expecting
5 to occur, you don't wait and let that happen for a
6 long period of time. You can implement actions
7 early to correct that. You don't wait until the
8 end of mining.

9 Another indication of a potential issue would
10 be if the groundwater inflows are much large than
11 you would expect, that would indicate that there
12 may be more drawdown. Don't expect that to happen
13 again but if it did, you could monitor for that and
14 you could correct that by implementing mitigation.

15 There are mitigation measures for these sorts
16 of things. There are mitigation measures such as
17 grouting that would reduce inflow to the wetlands.
18 And there are mitigation measures to reduce the
19 effects of drawdown if they occur. For instance,
20 adding water where that drawdown occurs.

21 With that, I'll let Cliff Twaroski come up and
22 talk more about the sulfate reduction.

23 MR. CLIFF TWAROSKI: My name is Cliff
24 Twaroski, T-W-A-R-O-S-K-I, and I'd like to continue
25 some discussions on sulfate reduction -- or

1 reducing sulfate and reducing methylmercury around
2 the project area.

3 One of the things struck our group was when
4 Dr. Branfireun presented this slide. And for me it
5 represented some information that I hadn't seen
6 before. There was the dosing study that
7 Dr. Branfireun mentioned at the Marcell
8 Experimental Forest. That was dosing with
9 kilograms per hectare per year of sulfate. And
10 Dr. Branfireun's information here is about
11 milligrams per liter of sulfate and that by adding
12 X amount of sulfate, there is a response in
13 methylmercury of 4X, 20X, and 30X. And so one of
14 the things that we are wondering about is that if
15 this is what happens when you add sulfate, what
16 happens when you take sulfate away?

17 And so taking the factor 30, because when we
18 look at reducing sulfate in the wastewater
19 treatment -- with the wastewater treatment system
20 from 200 milligrams to 10 milligrams per liter,
21 that's a reduction of about 190 milligrams per
22 liter. And based on the previous slide, we're not
23 sure where that really comes out on the scale that
24 Dr. Branfireun has presented.

25 But just as an example, we'll use that factor

1 of 30 to look at what happens when we take sulfate
2 away. And for Trimble Creek, which is identified
3 with this .7 ng/L concentration, we find that if
4 this can be applied, this laboratory study can be
5 applied to an actual environmental setting like
6 plant site and tailings basin, headwater wetlands,
7 there might be a reduction of over 1,000 grams per
8 year of methylmercury.

9 And if we look at Unnamed Creek headwater
10 wetland with a starting concentration of .4 ng/L,
11 based on these values, there could be about a 700
12 gram per year reduction in methylmercury.

13 I don't know if there is uncertainty certainly
14 associated with this, but it does provide some
15 perspective as to the potential magnitude of the
16 reductions that could occur with PolyMet's project
17 and operation and the seepage capture and treatment
18 removing sulfate from the system.

19 The other point that I want to talk about is
20 mass balance. We heard yesterday that mass balance
21 is a naive approach. I don't believe that's true.
22 Mass balance is informative. It can explain
23 watershed processes. It can help account for
24 things in the watershed. It's used by a number of
25 researchers, including Dr. Branfireun. And again,

1 that's what we have used to address various
2 questions that came up in front of the agencies.
3 The agency has a series of Ph.D. scientists that we
4 talk to about the best approach for using this in
5 support of the FEIS and in support of cross-media
6 analysis, and this is the approach that was
7 recommended.

8 And specifically for the cross-media analysis,
9 this was an analysis that was conducted to
10 specifically address the Band's concerns about
11 sulfide mineral dust adding sulfur to wetlands that
12 would then create more methylmercury to be flushed
13 downstream in the Partridge and Embarrass Rivers
14 downstream to the St. Louis River and the
15 Reservation.

16 That modeling used a combination -- or that
17 assessment used a combination of models, air
18 dispersion modeling with air model, which is a mass
19 balance model for air emissions. GoldSim was also
20 used for water flows. And where we had a gap in
21 how to address certain watershed functions,
22 including wetlands and what happens with the
23 sulfate and methylmercury, we then used mass
24 balance calculations to represent what might
25 happen.

1 That modeling confirmed that -- that
2 assessment confirmed that there would be a
3 reduction in mercury and sulfate and also
4 methylmercury. And we also assessed methylmercury
5 impacts to fish in the Embarrass and Partridge
6 Rivers. And at sites closest to the project we
7 found that there was no measurable change in fish
8 tissue mercury.

9 We did not use MPCA's mercury risk estimation
10 method. We used calculations and provided data
11 from the MPCA and Dr. Bruce Monson in calculation
12 concentration change and found that we did not have
13 a measurable change near the project. If we don't
14 have a measurable change near the project, it would
15 be very hard and almost -- and not likely that we
16 would ever see a change in fish mercury down the
17 St. Louis River.

18 That's the end of my presentation. Thank you.

19 MS. CHRISTIE KEARNEY: Thank you.
20 I'd just like to touch briefly again on adaptive
21 water management.

22 So adaptive water management is systematic
23 monitoring, modeling, and review process to improve
24 the performance of the project. This is used
25 around the world, recommended by the EPA and a lot

1 of processes and is a good way to be able to react
2 and respond to changes that you're seeing in your
3 project. It's a proactive approach that
4 anticipates uncertainty and variability by using
5 flexible and adaptive engineering controls like our
6 water treatment plant and establishes processes for
7 monitoring and responding to actual conditions as
8 they're occurring.

9 The permit to mine includes a condition
10 requiring an adaptive water management plan that's
11 designed such that adaptive management systems can
12 be implemented prior to reaching a water quality
13 limit. This is just one example of adaptive
14 management that is required by our permits.

15 The same plan is required by the NPDES permit
16 and our water appropriation permit. And this plan
17 is required to be reviewed and approved by the
18 agency before we can start our operation.

19 I also want to talk about uncertainty on
20 environmental outcomes since that's been discussed
21 quite a bit today and yesterday.

22 Certainty, absent certainty in environmental
23 predictions is a false goal. You will never have
24 absolute certainty. There is always uncertainty.
25 Therefore, you make reasonable and often

1 conservative estimates of outcomes based on your
2 data, your sound science, your engineering
3 principles, and peer and agency review.

4 So let's talk about modeling for a minute.
5 George Box, a well-respected statistician, once
6 said "all models are wrong but some are useful."
7 You use your data. You spend a lot of time in
8 calibration to try to match that natural system as
9 well as you can. You're never going to match a
10 natural system. Natural systems are complex.
11 There's a lot of different processes that happen,
12 and no model can match them directly. So we do
13 what we can and we spend time in calibration and
14 peer and agency review to try to get as close as we
15 possibly can.

16 The models that you saw us put up and the
17 models that you saw the Band put up have been
18 reviewed. They've been reviewed across the board
19 by agencies. They've been reviewed by each side of
20 this discussion today and yesterday. And the
21 models that have been put forward by our project
22 have been reviewed and approved by the agencies.
23 They've been run and rerun by the agencies. And
24 we've got our permits as a result.

25 We need to be conservative for immediate --

1 immediate critical risk items. So engineering
2 designs always have a factor of safety. You add a
3 factor of safety because there's uncertainty. In
4 everything we do there's uncertainty. You just
5 need to be able to make those reasonable
6 conservative assumptions to move forward.

7 And then you have adaptive management. You
8 need to identify a problem before a problem exists,
9 and you need to watch to make adjustments for it to
10 avoid the negative consequences such as a water
11 quality violation. You need to look for those
12 changes in water quality before they get to that
13 violation. You identify triggers where you're
14 going to take action to make changes.

15 Now, I'll turn it over to Jay.

16 MR. JAY JOHNSON: Good afternoon. My
17 name is Jay Johnson, J-O-H-N-S-O-N, and I am
18 outside legal counsel to PolyMet.

19 I had hoped that I would not speak at this
20 hearing which PolyMet believes should be focused on
21 science and facts. But having reviewed EPA's
22 recommendation and the Band's comments, I think
23 it's important to put some of PolyMet's legal views
24 on the hearing record.

25 To explain these views I'm going to use some

1 slides to walk through the text of Clean Water Act
2 Section 401(a) (2). Here is just the text of
3 401(a) (2). Fair to say it's a bit dense but we
4 will unpack it.

5 As we'll see, Section 401(a) (2) has six steps.
6 The first step, EPA is notified of a Section 401
7 certification and a permit application. That's
8 already happened.

9 Second step, EPA can determine that the
10 permitted discharges may affect the downstream
11 jurisdiction's water quality. That happened too.

12 The third step, the downstream jurisdiction,
13 in this case the Band, can determine that the
14 discharges will affect the quality of its waters so
15 as to 0 any water quality requirements and then
16 object to the permit.

17 We know this because in the text of the Act it
18 says exactly that. The Band made this
19 determination and sent in its "will affect" letter
20 in August of last year.

21 The fourth step. Federal permitting agency
22 holds a public hearing on objection where it hears
23 recommendations and evidence. That's where we are
24 right now. How do we know this happens? Again, we
25 look at the text. It says that the permitting

1 agency shall hold a hearing on the objection. That
2 means on the objection of the downstream
3 jurisdiction.

4 What's presented at the hearing? What's
5 presented at the hearing are the recommendations of
6 the State, in this case the Band, the EPA, and
7 additional evidence.

8 The next step is the most important one in our
9 view, but it's the step that both the Band and the
10 EPA skip over. The federal agency decides whether
11 the discharges that it has permitted will 0 the
12 downstream jurisdiction's water quality
13 requirements. Let's look at the text again. It
14 says that the permitting agency makes a decision
15 based upon the recommendations and evidence at the
16 hearing.

17 Notice too the decision is on the objection.
18 In other words, the agency decides, based on the
19 evidence presented at the hearing, should the
20 objection be sustained.

21 The next language is important too. The
22 permitting agency must condition the permit but
23 only as may be necessary to meet the downstream
24 standards. This "as may be necessary" language is
25 important because it means that condition may not

1 be necessary. If the permitting discharges won't 0
2 the downstream standards, conditions aren't
3 necessary. And if there is uncertainty, that
4 doesn't mean that there is a violation. It doesn't
5 mean the permitted discharges will 0 the downstream
6 water quality requirements.

7 Finally we have step six. If the permitted
8 discharges will 0 downstream water quality
9 requirements, the federal agency decides whether
10 additional permit conditions can ensure compliance
11 with those requirements. Let's take a look at the
12 text one more time.

13 This language tells us that the permitting
14 agency -- it tells the permitting agency to
15 condition the permit in a way that will ensure
16 compliance with water quality requirements. This
17 gives the permitting agency a lot of flexibility.
18 The EPA suggests in its recommendations that
19 monitoring and water quality requirements could
20 work but then it just dismisses them as
21 impractical. But as Christie just shared and as
22 we've said throughout our presentation, PolyMet is
23 already using many practical permit conditions to
24 meet water quality requirements. 7,000 permit
25 conditions total.

1 So in summary, the first issue in this hearing
2 and under Section 401(a) (2) is whether PolyMet's
3 permitted discharge violations will 0 the Band's
4 water quality requirements. No permit conditions
5 are necessary if there is no violation.

6 And the last thing I will bring up is the
7 question of burden. Section 401(a) (2) puts the
8 burden of persuasion on the party that is
9 objecting, the downstream jurisdiction. And this
10 hearing isn't happening on a clean slate.

11 PolyMet's project has been studied for 15 years.
12 The Corps and other co-lead agencies approved a
13 final EIS. The Corps issued the permit. So did
14 the Minnesota Pollution Control Agency. So did the
15 Minnesota Department of Natural Resources. The
16 Band is the one that's objecting and through this
17 hearing, seeking a different result. That means
18 the Band has the burden of persuasion.

19 Thank you. For listening. And now I'd like
20 to turn the floor over to PolyMet's chairman and
21 CEO John Cherry for the final word.

22 MR. JON CHERRY: Good afternoon. My
23 name is John Cherry. I'm the chairman, president,
24 and CEO of PolyMet Mining. It's J-O-N,
25 C-H-E-R-R-Y.

1 I'm also a registered professional engineer in
2 the field of environmental engineering, and I've
3 spent a little over 33 years in the mining
4 industry. The majority of that designing,
5 permitting, and building mines in the United
6 States.

7 You've heard about Eagle a few times today and
8 I'm pretty proud of that. I was the general
9 manager at Eagle. Did the design, permitting, and
10 litigation and construction of Eagle. And it's
11 turned out to be an incredibly successful mine in
12 many aspects, including the environmental
13 protections put in and designed into that.

14 The mine was originally started -- the
15 construction started in late 2010. Went through
16 2013 into early 2014 went into operation and is
17 still operating today.

18 During that entire period of time, it hasn't
19 had a single notice of violation of its
20 environmental permit conditions. I think that's a
21 testament to the design and the adaptive management
22 practices that were put in place at that operation.

23 Myself and some of the team that helped me at
24 Eagle are also here with me on the PolyMet project.
25 And we're taking many of the similar approaches and

1 things we used at Eagle that were so successful,
2 and we're implementing those on this project here.

3 I want to talk just for a second about real
4 life verses modeling. We've heard a lot about
5 modeling today. Christie touched on this and
6 others touched on this as well.

7 There's a certain level of certainty and
8 uncertainty in modeling. What I want to touch on
9 for a second are a couple of examples.

10 So when we did Eagle, obviously there were
11 project components that -- we were concerned about
12 the environment. They raised various concerns.
13 And they ran models. And one specific one that
14 I'll point out is Eagle is an underground mine.
15 And it -- the Salmon Trout River runs across the
16 top of the mine. There's corresponding wetland
17 corridor along the river across the top of the
18 mine.

19 GLIFWC and others ran a model. And their
20 model suggested that the river was going to dry up
21 and flow into the mine. The riparian wetlands were
22 going to dry up and be destroyed and there were
23 going to be thousands of gallons of water per
24 minute that would flow into the mine.

25 Our modeling estimate -- and I -- there's

1 always conservatism built into the assumptions of
2 the model. The modeling that our team and I put
3 together suggested that there would be a couple
4 hundred gallons per minute. So a couple of orders
5 of magnitude difference in those estimates about
6 those models.

7 So let's talk about what happened in real
8 life. In real life after the mine was built, there
9 wasn't a couple thousand gallons of water that went
10 in. There wasn't a couple hundred gallons per
11 minute of water that went in. The mine was so dry
12 they were pumping water into the mine for dust
13 control.

14 So these models are typically very
15 conservative and intentionally so to protect the
16 environment. Make sure we get it right. But when
17 you build conservatism upon conservatism upon
18 conservatism, sometimes you end up with
19 overestimates. And that's typically what happens
20 in these type of projects.

21 Other criticisms at Eagle by the opponents
22 were that the treatment system wouldn't work. It
23 wasn't a proven treatment technology. It was
24 proven. Membrane technology has been around for a
25 long time. People have been using that for a long

1 time to make drinking water. It's also the
2 technology that can knock mercury out. It can get
3 sulfate out. In our case we have to use membrane
4 technology to get the sulfate down to the 10
5 standard, 10 mg/L because that's what we committed
6 to do. We're the only mining company that's agreed
7 to meet that 10 sulfate standard for wild rice
8 protection. And the membrane technology can do
9 that.

10 So what we did -- so there's the treatment
11 technology and there's also cutoff. We said the
12 deep cutoff wall of 85. They said that would not
13 work. That was installed. It worked and has been
14 demonstrated. And again, Eagle has gone through
15 all these years from about 2013 or '14 through
16 current, as far as I know, without a single notice
17 of violation to their environmental permits. So
18 the technology is there. Can be done and can be
19 done successfully. Technology has been proven.

20 So we've -- myself and my team and I, we've
21 taken what we've learned, taken best of what we saw
22 at Eagle and we tried to implement some of those
23 best practices at PolyMet.

24 Part of the reason I'm at PolyMet -- I've been
25 here about 10 years now. And shortly after I did

1 Eagle, I left there and worked on some other mining
2 projects. But back in about 2010 or 2012 time
3 period, you heard about EIS that was described
4 yesterday where we got the EU3 rating, got the
5 failed rating on that.

6 Shortly after that, I was approached and asked
7 to join PolyMet, see if I can take what we learned
8 at Eagle and apply it to PolyMet and make it
9 better. And what I'd like to say is that the
10 process worked. There was a public comment period.
11 Everyone participated. Many people were listened
12 to. And the project changed as a result of the
13 project for the better.

14 Those changes to the project resulted in an
15 EC2 rating, as Christie mentioned. And this is
16 something I'm very proud of and I know our team is
17 proud of. It's the highest rating that a mining
18 project has ever received in the United States from
19 the EPA. We're very proud of that fact. That told
20 me we were on the right track and we were doing the
21 right thing.

22 And it's -- I'm sure it was an oversight
23 yesterday. We heard a lot from the Band about the
24 failed EIS and the original EIS, but they never
25 mentioned the improvements to the project in the

1 EC2 rating that the EPA provided.

2 Just last couple things I want to touch on as
3 we wrap up here, and this was touched on earlier
4 this morning but I think this is very, very
5 important.

6 Last June the White House issued their report
7 on critical minerals in the United states. On
8 page 99 of the report it specifically noted that
9 PolyMet would be the only nickel mine in the
10 country when Eagle shuts down in 2025. Nickel is a
11 critical mineral needed for batteries, needed for
12 the green economy. And we don't want -- the
13 majority of the nickel in the world right now comes
14 out of Russia. Not a great place to be getting
15 nickel from. So there's a strategic reason why we
16 need to develop these resources in the United
17 States and we have that. We have a permitted
18 project ready to go and produce that nickel right
19 here as well as cobalt and some other strategic
20 metals.

21 Then just a few weeks ago President Biden
22 issued a directive under the Defense Production Act
23 about how important it is to develop these
24 strategic metals and minerals here in the United
25 States for our own strategic defense purposes.

1 Like I said, we've got a project that's
2 permitted. We're ready to go. We think we can do
3 it. I wouldn't -- I wouldn't stand here in front
4 of you and promote this project and put it forward
5 if I didn't think that we could do it and do it
6 safely.

7 Protecting the environment, cleaning up the
8 environment, and producing these metals, it's not a
9 mutually exclusive proposition. We can do both.
10 And our team, especially the team that lives at the
11 site, they live there, they play there, they
12 recreate there, they raise their families there. I
13 can't think of anyone that has more vested interest
14 in doing this correctly and doing it right and
15 protecting the environment than our team that lives
16 there, and they're the ones that will ensure that
17 it's done correctly.

18 So in conclusion, Colonel Jansen, I want to
19 thank you for your time and your team's time. I
20 thought this was an excellent hearing. A great
21 facility. Very well run, very efficient hearing
22 today. Appreciate the attention and listening to
23 what we have to say about the project.

24 And I want to thank our team specifically for
25 all the hours and hard work they put into the

1 presentation and sharing our story here. It's a
2 great story. And if you can't tell, I'm very
3 passionate and proud of what we have.

4 After 15 years, it's time to move this project
5 forward. It's been studied to death. It's been
6 looked at upside down, sideways, and from every
7 direction. I think it holds up and I think we're
8 ready to go. And we very respectfully ask you to
9 consider what we've presented and to reinstate the
10 permit as quickly as we can.

11 Thank you.

12 COLONEL JANSEN: Thank you, Christie,
13 Steve, Greg, Cliff, Jay, and Jon for your
14 statements. This concludes Day 2 of the public
15 hearing regarding Fond du Lac's objection to the
16 Corps' Section 404 permit for the PolyMet NorthMet
17 Mine project.

18 Thank you to all the parties representing Fond
19 du Lac, USEPA, PolyMet. To all of our witnesses I
20 personally thank you for your preparation and
21 presentation of valuable information and for
22 sharing your social, cultural, legal, and
23 scientific perspectives over the last two days.

24 Also, a big thank you to all of those who
25 planned and facilitated the hearing here at the

1 resort.

2 We'll reconvene tomorrow afternoon at 4 p.m.
3 for Day 3 of this public hearing. Tomorrow we'll
4 receive verbal statements from the public via
5 teleconference. And information on the call in for
6 tomorrow is available on the Corps' PolyMet project
7 web page. That link is also on the slide displayed
8 here.

9 So again, I want to thank all of you, wish you
10 safe travels, and a good afternoon.

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1 STATE OF MINNESOTA)
) ss.
2 COUNTY OF WASHINGTON)

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BE IT KNOWN, that I took the proceedings at the time and place set forth herein;

That the proceedings were recorded in shorthand and transcribed into typewriting, that the transcript is a true record of the proceedings, to the best of my ability;

That I am not related to any of the parties hereto nor interested in the outcome of the action;

IN EVIDENCE HEREOF, WITNESS MY HAND AND SEAL.

Lisa M. Thorsgaard